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5,15-Bis(4-pentyloxyphenyl)porphyrin

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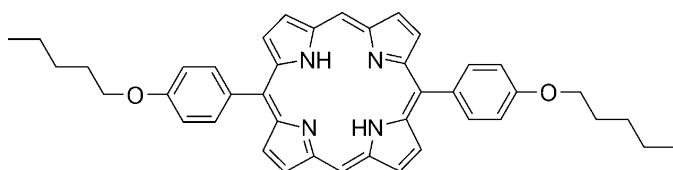
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Key indicators: single-crystal X-ray study; $T = 90$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.045; wR factor = 0.109; data-to-parameter ratio = 16.5.

In the title compound, $\text{C}_{42}\text{H}_{42}\text{N}_4\text{O}_2$, the complete molecule is generated by a crystallographic inversion centre. The porphyrin system exhibits a near planar macrocycle conformation with an average deviation from the least-squares plane of the 24 macrocycle atoms of 0.037 (5) Å. The phenyl *ipso* C atoms are positioned above and below the porphyrin plane by 0.35 (1) Å and the macrocycle shows evidence of in-plane rectangular elongation with $\text{N}\cdots\text{N}$ separations of 3.032 (5) and 2.803 (5) Å. Two intramolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds occur.

Related literature

For the conformation of porphyrins, see: Scheidt & Lee (1987); Senge *et al.* (1997); Senge (2006). For the synthesis of such compounds, see: Wiehe *et al.* (2005).



Experimental

Crystal data

$\text{C}_{42}\text{H}_{42}\text{N}_4\text{O}_2$
 $M_r = 634.80$
Triclinic, $P\bar{1}$

$a = 9.5222$ (6) Å
 $b = 9.5799$ (6) Å
 $c = 10.2195$ (6) Å

$\alpha = 67.777$ (1)°
 $\beta = 88.063$ (1)°
 $\gamma = 72.464$ (1)°
 $V = 819.49$ (9) Å³
 $Z = 1$

Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 90$ K
 $0.30 \times 0.10 \times 0.08$ mm

Data collection

Bruker SMART APEXII diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2005)
 $T_{\min} = 0.97$, $T_{\max} = 0.99$

9093 measured reflections
3606 independent reflections
2489 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.109$
 $S = 1.04$
3606 reflections

219 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.27$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.23$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N21}-\text{H21}\cdots\text{N24}$	0.88	2.50	3.033 (2)	119
$\text{N21}-\text{H21}\cdots\text{N24}^i$	0.88	2.22	2.804 (2)	123

Symmetry code: (i) $-x + 1, -y, -z + 1$.

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: XP in SHELXTL (Sheldrick, 2008); software used to prepare material for publication: XP in SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZL2552).

References

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Wiehe, A., Shaker, Y. M., Brandt, J. C., Mebs, S. & Senge, M. O. (2005). *Tetrahedron*, **61**, 5535–5564.

supplementary materials

Acta Cryst. (2013). E69, o1048 [doi:10.1107/S160053681301550X]

5,15-Bis(4-pentyloxyphenyl)porphyrin**Mathias O. Senge****Comment**

Many porphyrin structures with four *meso* substituents have been reported (Scheidt & Lee, 1987). The available number of structures for systems with only two *meso* residues is much smaller. In the context of an ongoing program on the conformational flexibility of porphyrins (Senge, 2006) we are interested in a comparative analysis of 5,10- A_2 - and 5,15- A_2 -disubstituted porphyrins. The title compound is an example for the latter and exhibits a planar macrocycle with an average deviation from the least-squares-plane of the 24 macrocycle atoms of $\Delta_{24} = 0.037(5)$ Å. The phenyl *ipso* carbon atoms are positioned above and below the porphyrin plane by 0.35 Å and the macrocycle shows evidence for in-plane distortion with N··N separations of 3.032(5) and 2.803(5) Å. This is similar to the situation found in 2,3,5,7,8,12,13,15,17,18- decasubstituted porphyrins (Senge *et al.*, 1997) where *peri* interaction between the *meso* and beta substituents occur. The molecules pack in parallel layers with the alkyl chains separating the macrocycles and only minimal π -aggregation.

Experimental

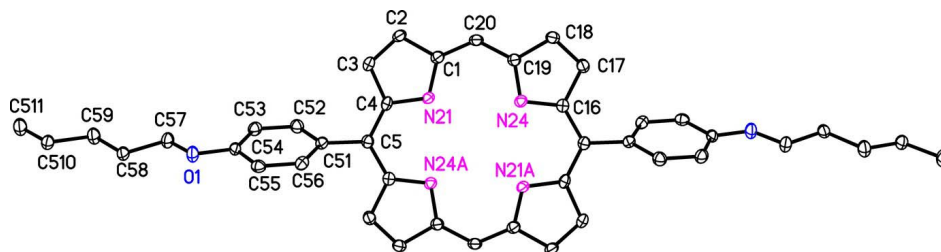
The compound was prepared as described by Wiehe *et al.* (2005) and crystallized from CH_2Cl_2/CH_3OH .

Refinement

All nonhydrogen atoms were refined with anisotropic thermal parameters. Hydrogen atoms were refined with a standard riding model (C—H distance 0.96 - 0.99 Å, $U_{iso} = 1.2$ –1.5 times of parent atom). Pyrrole hydrogen atoms were located in difference maps and refined with isotropic thermal parameters.

Computing details

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINTE* (Bruker, 2005); data reduction: *SAINTE* (Bruker, 2005); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *XP* in *SHELXTL* (Sheldrick, 2008).


Figure 1

Molecular structure of the title compound. Thermal ellipsoids are drawn at 50% probability level; hydrogen atoms have been omitted for clarity.

5,15-Bis(4-pentyloxyphenyl)porphyrin

Crystal data

$C_{42}H_{42}N_4O_2$

$M_r = 634.80$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 9.5222\ (6)\ \text{\AA}$

$b = 9.5799\ (6)\ \text{\AA}$

$c = 10.2195\ (6)\ \text{\AA}$

$\alpha = 67.777\ (1)^\circ$

$\beta = 88.063\ (1)^\circ$

$\gamma = 72.464\ (1)^\circ$

$V = 819.49\ (9)\ \text{\AA}^3$

$Z = 1$

$F(000) = 338$

$D_x = 1.286\ \text{Mg m}^{-3}$

$D_m = n/d\ \text{Mg m}^{-3}$

D_m measured by not measured

Melting point: n/d K

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 1771 reflections

$\theta = 4.5\text{--}60.7^\circ$

$\mu = 0.08\ \text{mm}^{-1}$

$T = 90\ \text{K}$

Parallelepiped, red

$0.30 \times 0.10 \times 0.08\ \text{mm}$

Data collection

Bruker SMART APEXII
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: $8.3\ \text{pixels mm}^{-1}$

ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2005)

$T_{\min} = 0.97$, $T_{\max} = 0.99$

9093 measured reflections

3606 independent reflections

2489 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.039$

$\theta_{\max} = 27.1^\circ$, $\theta_{\min} = 2.2^\circ$

$h = -12 \rightarrow 12$

$k = -12 \rightarrow 12$

$l = -13 \rightarrow 13$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.045$

$wR(F^2) = 0.109$

$S = 1.04$

3606 reflections

219 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0433P)^2 + 0.1316P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.27\ \text{e \AA}^{-3}$

$\Delta\rho_{\min} = -0.23\ \text{e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N21	0.64309 (14)	-0.19503 (16)	0.46008 (14)	0.0171 (3)
H21	0.5815	-0.1069	0.4627	0.047 (6)*
N24	0.36841 (13)	-0.12143 (15)	0.61217 (14)	0.0160 (3)
C1	0.62390 (16)	-0.34160 (19)	0.52069 (17)	0.0179 (4)
C2	0.74472 (17)	-0.4492 (2)	0.48647 (18)	0.0190 (4)
H2A	0.7600	-0.5592	0.5123	0.023*
C3	0.83431 (17)	-0.36651 (19)	0.41037 (17)	0.0185 (4)
H3A	0.9229	-0.4090	0.3740	0.022*
C4	0.77200 (16)	-0.20511 (19)	0.39479 (17)	0.0169 (4)
C5	0.83034 (16)	-0.07952 (19)	0.33369 (17)	0.0169 (4)
C16	0.24129 (16)	-0.07563 (19)	0.67302 (16)	0.0164 (3)
C17	0.18961 (17)	-0.2107 (2)	0.74985 (18)	0.0195 (4)
H17A	0.1041	-0.2097	0.8006	0.023*
C18	0.28721 (17)	-0.3376 (2)	0.73457 (18)	0.0198 (4)
H18A	0.2843	-0.4437	0.7732	0.024*
C19	0.39727 (16)	-0.28065 (19)	0.64781 (17)	0.0172 (4)
C20	0.51314 (16)	-0.37959 (19)	0.60556 (17)	0.0185 (4)
H20A	0.5167	-0.4877	0.6396	0.022*
C51	0.98127 (17)	-0.11283 (19)	0.28262 (17)	0.0177 (4)
C52	1.01657 (17)	-0.1671 (2)	0.17315 (18)	0.0206 (4)
H52A	0.9421	-0.1859	0.1286	0.025*
C53	1.15769 (17)	-0.1945 (2)	0.12745 (18)	0.0207 (4)
H53A	1.1792	-0.2316	0.0527	0.025*
C54	1.26742 (16)	-0.16720 (19)	0.19238 (18)	0.0188 (4)
C55	1.23542 (17)	-0.11454 (19)	0.30202 (17)	0.0192 (4)
H55A	1.3103	-0.0965	0.3467	0.023*
C56	1.09475 (17)	-0.08822 (19)	0.34664 (18)	0.0192 (4)
H56A	1.0744	-0.0526	0.4224	0.023*
O1	1.41004 (11)	-0.18754 (14)	0.15628 (12)	0.0226 (3)
C57	1.45008 (17)	-0.2298 (2)	0.03660 (18)	0.0227 (4)
H57A	1.3832	-0.1524	-0.0488	0.027*
H57B	1.4430	-0.3366	0.0554	0.027*
C58	1.60744 (17)	-0.2284 (2)	0.01421 (18)	0.0211 (4)
H58A	1.6702	-0.2943	0.1048	0.025*
H58B	1.6106	-0.1185	-0.0147	0.025*
C59	1.66924 (17)	-0.2919 (2)	-0.09922 (19)	0.0249 (4)
H59A	1.6492	-0.3934	-0.0788	0.030*

H59B	1.6167	-0.2157	-0.1927	0.030*
C510	1.83383 (17)	-0.3191 (2)	-0.10777 (18)	0.0217 (4)
H51A	1.8868	-0.3995	-0.0159	0.026*
H51B	1.8546	-0.2190	-0.1236	0.026*
C511	1.89275 (18)	-0.3744 (2)	-0.22522 (19)	0.0277 (4)
H51C	2.0009	-0.4029	-0.2182	0.042*
H51D	1.8518	-0.2889	-0.3175	0.042*
H51E	1.8637	-0.4673	-0.2159	0.042*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N21	0.0139 (7)	0.0134 (7)	0.0229 (8)	-0.0023 (5)	0.0014 (5)	-0.0075 (6)
N24	0.0136 (6)	0.0145 (7)	0.0195 (7)	-0.0030 (5)	-0.0003 (5)	-0.0071 (6)
C1	0.0159 (8)	0.0165 (8)	0.0220 (9)	-0.0040 (6)	-0.0007 (6)	-0.0088 (7)
C2	0.0184 (8)	0.0141 (8)	0.0241 (9)	-0.0023 (6)	0.0006 (7)	-0.0090 (7)
C3	0.0148 (8)	0.0197 (9)	0.0206 (9)	-0.0017 (6)	0.0021 (6)	-0.0103 (7)
C4	0.0119 (7)	0.0192 (9)	0.0184 (9)	-0.0012 (6)	0.0004 (6)	-0.0087 (7)
C5	0.0151 (8)	0.0190 (9)	0.0171 (8)	-0.0047 (6)	0.0008 (6)	-0.0078 (7)
C16	0.0156 (8)	0.0186 (9)	0.0146 (8)	-0.0046 (6)	0.0001 (6)	-0.0065 (7)
C17	0.0179 (8)	0.0206 (9)	0.0199 (9)	-0.0071 (7)	0.0040 (6)	-0.0071 (7)
C18	0.0193 (8)	0.0156 (9)	0.0237 (9)	-0.0066 (7)	0.0022 (7)	-0.0059 (7)
C19	0.0153 (8)	0.0159 (8)	0.0201 (9)	-0.0046 (6)	-0.0004 (6)	-0.0066 (7)
C20	0.0180 (8)	0.0136 (8)	0.0236 (9)	-0.0054 (6)	-0.0001 (6)	-0.0063 (7)
C51	0.0171 (8)	0.0144 (8)	0.0191 (9)	-0.0031 (6)	0.0028 (6)	-0.0052 (7)
C52	0.0183 (8)	0.0214 (9)	0.0221 (9)	-0.0066 (7)	0.0012 (7)	-0.0080 (8)
C53	0.0232 (9)	0.0212 (9)	0.0189 (9)	-0.0060 (7)	0.0045 (7)	-0.0100 (7)
C54	0.0146 (8)	0.0146 (8)	0.0222 (9)	-0.0026 (6)	0.0047 (6)	-0.0036 (7)
C55	0.0193 (8)	0.0165 (9)	0.0204 (9)	-0.0057 (7)	-0.0015 (7)	-0.0053 (7)
C56	0.0197 (8)	0.0182 (9)	0.0184 (9)	-0.0034 (7)	0.0021 (6)	-0.0078 (7)
O1	0.0168 (6)	0.0286 (7)	0.0251 (7)	-0.0071 (5)	0.0065 (5)	-0.0135 (6)
C57	0.0202 (8)	0.0268 (10)	0.0229 (9)	-0.0062 (7)	0.0069 (7)	-0.0127 (8)
C58	0.0180 (8)	0.0183 (9)	0.0233 (9)	-0.0047 (7)	0.0060 (7)	-0.0050 (7)
C59	0.0204 (9)	0.0312 (11)	0.0260 (10)	-0.0096 (8)	0.0071 (7)	-0.0135 (8)
C510	0.0185 (8)	0.0216 (9)	0.0237 (9)	-0.0046 (7)	0.0049 (7)	-0.0088 (8)
C511	0.0209 (9)	0.0360 (11)	0.0324 (11)	-0.0105 (8)	0.0071 (8)	-0.0190 (9)

Geometric parameters (\AA , $^\circ$)

N21—C1	1.370 (2)	C52—H52A	0.9500
N21—C4	1.3722 (19)	C53—C54	1.394 (2)
N21—H21	0.8800	C53—H53A	0.9500
N24—C19	1.367 (2)	C54—O1	1.3710 (18)
N24—C16	1.3711 (19)	C54—C55	1.384 (2)
C1—C20	1.388 (2)	C55—C56	1.381 (2)
C1—C2	1.428 (2)	C55—H55A	0.9500
C2—C3	1.362 (2)	C56—H56A	0.9500
C2—H2A	0.9500	O1—C57	1.4335 (19)
C3—C4	1.427 (2)	C57—C58	1.511 (2)
C3—H3A	0.9500	C57—H57A	0.9900

C4—C5	1.399 (2)	C57—H57B	0.9900
C5—C16 ⁱ	1.412 (2)	C58—C59	1.527 (2)
C5—C51	1.496 (2)	C58—H58A	0.9900
C16—C5 ⁱ	1.412 (2)	C58—H58B	0.9900
C16—C17	1.457 (2)	C59—C510	1.515 (2)
C17—C18	1.348 (2)	C59—H59A	0.9900
C17—H17A	0.9500	C59—H59B	0.9900
C18—C19	1.448 (2)	C510—C511	1.514 (2)
C18—H18A	0.9500	C510—H51A	0.9900
C19—C20	1.396 (2)	C510—H51B	0.9900
C20—H20A	0.9500	C511—H51C	0.9800
C51—C52	1.394 (2)	C511—H51D	0.9800
C51—C56	1.403 (2)	C511—H51E	0.9800
C52—C53	1.389 (2)		
C1—N21—C4	110.37 (13)	C54—C53—H53A	120.3
C1—N21—H21	124.8	O1—C54—C55	114.91 (14)
C4—N21—H21	124.8	O1—C54—C53	125.17 (15)
C19—N24—C16	105.16 (13)	C55—C54—C53	119.92 (14)
N21—C1—C20	126.59 (15)	C56—C55—C54	120.09 (15)
N21—C1—C2	106.65 (13)	C56—C55—H55A	120.0
C20—C1—C2	126.68 (15)	C54—C55—H55A	120.0
C3—C2—C1	108.13 (15)	C55—C56—C51	121.49 (16)
C3—C2—H2A	125.9	C55—C56—H56A	119.3
C1—C2—H2A	125.9	C51—C56—H56A	119.3
C2—C3—C4	108.17 (14)	C54—O1—C57	118.76 (12)
C2—C3—H3A	125.9	O1—C57—C58	106.88 (13)
C4—C3—H3A	125.9	O1—C57—H57A	110.3
N21—C4—C5	124.67 (14)	C58—C57—H57A	110.3
N21—C4—C3	106.62 (14)	O1—C57—H57B	110.3
C5—C4—C3	128.61 (14)	C58—C57—H57B	110.3
C4—C5—C16 ⁱ	123.29 (14)	H57A—C57—H57B	108.6
C4—C5—C51	118.72 (14)	C57—C58—C59	111.63 (14)
C16 ⁱ —C5—C51	117.82 (14)	C57—C58—H58A	109.3
N24—C16—C5 ⁱ	125.69 (15)	C59—C58—H58A	109.3
N24—C16—C17	110.71 (14)	C57—C58—H58B	109.3
C5 ⁱ —C16—C17	123.59 (14)	C59—C58—H58B	109.3
C18—C17—C16	106.36 (14)	H58A—C58—H58B	108.0
C18—C17—H17A	126.8	C510—C59—C58	113.42 (14)
C16—C17—H17A	126.8	C510—C59—H59A	108.9
C17—C18—C19	106.69 (15)	C58—C59—H59A	108.9
C17—C18—H18A	126.7	C510—C59—H59B	108.9
C19—C18—H18A	126.7	C58—C59—H59B	108.9
N24—C19—C20	126.46 (14)	H59A—C59—H59B	107.7
N24—C19—C18	111.07 (13)	C59—C510—C511	113.04 (14)
C20—C19—C18	122.44 (15)	C59—C510—H51A	109.0
C1—C20—C19	128.88 (15)	C511—C510—H51A	109.0
C1—C20—H20A	115.6	C59—C510—H51B	109.0
C19—C20—H20A	115.6	C511—C510—H51B	109.0

C52—C51—C56	117.34 (14)	H51A—C510—H51B	107.8
C52—C51—C5	123.54 (14)	C510—C511—H51C	109.5
C56—C51—C5	119.12 (15)	C510—C511—H51D	109.5
C53—C52—C51	121.78 (15)	H51C—C511—H51D	109.5
C53—C52—H52A	119.1	C510—C511—H51E	109.5
C51—C52—H52A	119.1	H51C—C511—H51E	109.5
C52—C53—C54	119.38 (16)	H51D—C511—H51E	109.5
C52—C53—H53A	120.3		
C4—N21—C1—C20	174.18 (15)	C2—C1—C20—C19	176.51 (16)
C4—N21—C1—C2	-2.51 (18)	N24—C19—C20—C1	0.4 (3)
N21—C1—C2—C3	1.54 (18)	C18—C19—C20—C1	178.43 (16)
C20—C1—C2—C3	-175.14 (16)	C4—C5—C51—C52	63.4 (2)
C1—C2—C3—C4	-0.05 (19)	C16 ⁱ —C5—C51—C52	-121.15 (18)
C1—N21—C4—C5	-174.17 (15)	C4—C5—C51—C56	-116.84 (18)
C1—N21—C4—C3	2.48 (18)	C16 ⁱ —C5—C51—C56	58.6 (2)
C2—C3—C4—N21	-1.45 (18)	C56—C51—C52—C53	-0.7 (2)
C2—C3—C4—C5	175.01 (16)	C5—C51—C52—C53	179.07 (15)
N21—C4—C5—C16 ⁱ	-2.8 (3)	C51—C52—C53—C54	0.0 (2)
C3—C4—C5—C16 ⁱ	-178.73 (16)	C52—C53—C54—O1	-179.13 (15)
N21—C4—C5—C51	172.32 (14)	C52—C53—C54—C55	0.5 (2)
C3—C4—C5—C51	-3.6 (3)	O1—C54—C55—C56	179.34 (14)
C19—N24—C16—C5 ⁱ	179.15 (15)	C53—C54—C55—C56	-0.4 (2)
C19—N24—C16—C17	0.37 (17)	C54—C55—C56—C51	-0.4 (2)
N24—C16—C17—C18	0.14 (18)	C52—C51—C56—C55	0.9 (2)
C5 ⁱ —C16—C17—C18	-178.68 (15)	C5—C51—C56—C55	-178.90 (15)
C16—C17—C18—C19	-0.56 (18)	C55—C54—O1—C57	-175.16 (14)
C16—N24—C19—C20	177.47 (16)	C53—C54—O1—C57	4.5 (2)
C16—N24—C19—C18	-0.72 (17)	C54—O1—C57—C58	174.70 (13)
C17—C18—C19—N24	0.84 (19)	O1—C57—C58—C59	172.84 (13)
C17—C18—C19—C20	-177.45 (15)	C57—C58—C59—C510	-170.19 (15)
N21—C1—C20—C19	0.5 (3)	C58—C59—C510—C511	-177.21 (15)

Symmetry code: (i) $-x+1, -y, -z+1$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N21—H21 \cdots N24	0.88	2.50	3.033 (2)	119
N21—H21 \cdots N24 ⁱ	0.88	2.22	2.804 (2)	123

Symmetry code: (i) $-x+1, -y, -z+1$.