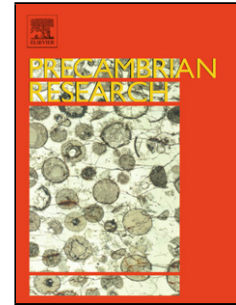


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Highlights

- black shale of the rift-related late Mesoproterozoic Arctic Bay Fm. was deposited under a stratified oxygenated / anoxic water column
- metal enrichments were caused primarily by stripping from the water column and possibly enhanced by diagenetic redistribution into subtle carbonate-bearing layers
- concentrated venting was absent at the study location, but basin geochemistry was suitable for formation of SEDEX and polymetallic deposits
- U-Th-Pb whole-rock depositional age of the Arctic Bay Formation is 1092 ± 59 Ma

1 **Arctic Bay Formation, Borden Basin, Nunavut (Canada): basin**
2 **evolution, black shale, and dissolved metal systematics in the**
3 **Mesoproterozoic ocean**

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18 **ABSTRACT**

19 The Arctic Bay Formation (Nunavut, Canada) represents a late Mesoproterozoic muddy
20 terrigenous ramp and contains >200 m of black shale. The formation was studied in order to
21 decipher the tectonostratigraphic and geochemical evolution of the basin, address the origin of
22 metal enrichment, and determine whether this frontier basin has the potential to host
23 sedimentary-exhalative or polymetallic black shale deposits. Samples were analysed in the
24 laboratory for major and trace elements, total organic carbon (TOC), 4-step loss-on-ignition
25 (LOI), and Pb isotopes. Non-calcareous black shale exhibits neither Ce nor Y anomalies,
26 reflecting euxinia in the lower water column, whereas slightly dolomitic black shale has both Ce
27 and Y anomalies, reflecting the dolomite's probable origin as a precipitate in the upper water
28 column. The stratigraphic distribution of the rare earth elements (REEs) indicates an evolving
29 sediment provenance, and Pb isotopic data indicate that the source of clay in the black shale was
30 dominated by weathered, juvenile, mantle-derived material. Base metals and redox-sensitive
31 metals, expressed as enrichment ratios relative to conservative lithophile elements, are elevated
32 and exhibit coherent covariations in the black shale. Enrichment in the redox-sensitive elements,
33 such as Mo and U, correlates with dolomite content of the shale, rather than with organic C or
34 Fe_{py} . From a deep-time ocean evolution perspective, this important observation suggests that
35 enrichment in these metals cannot necessarily be attributed to metal incorporation at an interface
36 between sediment and euxinic water. Instead, in Arctic Bay Formation black shale, the metals
37 were either scavenged onto dolomite as it precipitated in the water column, or secondarily re-
38 distributed within the sediment according to its dolomite content. The base metals that are
39 concentrated in the black shale (e.g., Zn) were probably sourced from diffuse hydrothermal
40 venting, and although there is no evidence at the studied location for a nearby point source of

41 metals (vent), persistent bottom- water euxinia would have ensured the effective scavenging of
42 any dissolved metals supplied, and so the basin has at the very minimum a hypothetical potential
43 for SEDEX and polymetallic mineralisation. Whole-rock U-Th-Pb isotope analysis of black
44 shale yielded a date of 1092 ± 59 Ma, which is considered to be the Arctic Bay Formation's
45 depositional age.

46 Keywords: black shale; geochemistry; Mesoproterozoic; Arctic Bay Formation; Borden Basin

47

48 **1. INTRODUCTION**

49 The Mesoproterozoic Borden Basin of the Canadian Arctic islands is an established
50 mining district (carbonate-hosted Nanisivik deposit), but its tectonostratigraphic evolution is
51 poorly understood, the basin's metal potential is underexplored, and many of its stratigraphic
52 units have received only cursory attention. One of the stratigraphic units in this rift basin, the >1
53 km-thick Arctic Bay Formation, contains an impressive thickness of black shale about which
54 little is known. Stratigraphic and geochemical data from the Arctic Bay Formation are used in
55 this study as the basis for a discussion of the basin's tectonic and geochemical evolution, the
56 distribution of metals as a function of stratigraphy, and the dynamics of dissolved redox-sensitive
57 metals in Mesoproterozoic seawater. Such elements are critical to understanding the evolution of
58 the Proterozoic atmosphere and ocean, but also provide a means of identifying possible horizons
59 of economic interest in black shale basins. The Mesoproterozoic represents a global nadir in the
60 temporal distribution of SEDEX deposits (Leach et al., 2010), a phenomenon that may be linked
61 to the still poorly understood geochemical evolution of Earth's surface environments at the time.

62

63 2 .GEOLOGICAL SETTING

64 The Bylot Supergroup, a ~6 km-thick succession of Mesoproterozoic strata, was
65 deposited in the Borden Basin, one of several ~1.2 Ga basins in the Canadian Arctic islands and
66 northwestern Greenland (Figs. 1, 2A; Fahrig et al., 1981; Jackson and Iannelli, 1981). The
67 Borden Basin consists of three “troughs” (Fig. 1), of which the Milne Inlet Graben (MIG) is both
68 the largest, and the only one known to contain sulphide deposits (Nanisivik ore-body and other
69 carbonate-hosted showings; Turner, 2011). The present study addresses the Arctic Bay
70 Formation, a thick succession of fine-grained terrigenous clastic rocks in the lower Bylot
71 Supergroup (Fig. 2). Strata of the basal Bylot Supergroup in the MIG locally reach prehnite-
72 pumpellyite to greenschist facies, but strata higher in the succession remain unmetamorphosed
73 (Jackson and Morgan, 1978; Galley et al., 1983; Dostal et al., 1989).

74 The MIG is a gently west-plunging trough with an abrupt northern margin and a more
75 gradual southern margin (Fig. 1). The regional structure of the MIG (Jackson and Iannelli, 1981;
76 Scott and deKemp, 1998) is dominated by pronounced northwest-trending intragaben and
77 graben-margin fault systems. These faults accommodated both early extension and continued
78 tectonic adjustments, strongly affected sedimentary facies distribution throughout MIG history,
79 were the locus of deep-water carbonate mound growth of the Ikpiarjuk Formation (Turner,
80 2009), and were intermittently reactivated during the Phanerozoic (Jackson and Iannelli, 1981).

81 Rifting of the Archean basement gneiss (Rae Province) is recorded by a basal subaerial
82 tholeiitic basalt (Nauyat Formation) that is related to the ~1.27 Ga Mackenzie igneous province
83 (Fig. 2A; LeCheminant and Heaman, 1989). The basalt is overlain successively by the
84 predominantly shallow-marine Adams Sound Formation quartz arenite and shale of the Arctic

85 Bay Formation (Jackson and Iannelli, 1981; Long and Turner, in press). The shale-dominated
86 Arctic Bay Formation was described by Jackson and Iannelli (1981) as thinning northwestward,
87 from ~770 m in most of the eastern Milne Inlet Graben to 180 m at Arctic Bay. The basal,
88 gradational contact with the underlying Adams Sound Formation quartz arenite was interpreted
89 as conformable. The upper contacts, with whichever of the three carbonate formations overlies it
90 in any given location, are also conformable (Turner, 2009). Two paleogeographic zones were
91 identified in the Arctic Bay Formation, with a boundary roughly at Milne Inlet (Jackson and
92 Iannelli, 1981). The Arctic Bay Formation shallows upward in the southeastern MIG, where it is
93 conformably overlain by the mixed carbonate-siliciclastic Iqqittuq Formation (Turner, 2009)
94 which contains sulphate-facies evaporite beds (Jackson and Cumming, 1981; Kah et al., 2001).
95 The Iqqittuq Formation is a northwest-deepening ramp that passes laterally to time-equivalent
96 deep-water shale of the upper Arctic Bay Formation west of Tremblay Sound (Figs. 1, 2A). In
97 the northwestern MIG, Arctic Bay Formation sandstone, siltstone and grey shale deepens upward
98 into black shale that is locally laterally equivalent to large, deep-water carbonate mounds
99 (Ikpiarjuk Formation) that accumulated in the vicinity of intragaben faults (Figs. 1, 2A,B,C).
100 The upper Arctic Bay Formation black shale is, therefore, geometrically and temporally
101 equivalent to both the Iqqittuq Formation and at least part of the Ikpiarjuk Formation (Fig. 2A).
102 In the northwestern MIG, the Arctic Bay Formation is recessive and poorly exposed, with the
103 exception of several locations where relatively continuous exposure is present in steep gullies
104 and canyons. This formation has received no formal study since the work of Jackson and Iannelli
105 (1981).

106 The Iqqittuq Formation carbonate ramp of the southeastern MIG is gradationally overlain
107 by cyclic, peritidal facies of the Angmaat Formation (rimmed platform). This thick platformal

108 unit passes westward over a short distance near Tremblay Sound to deep-water carbonate
109 laminite of the Nanisivik Formation, which hosts most of the Zn-Pb showings in the MIG
110 (Turner, 2009, 2011). The Nanisivik and Angmaat formations are separated from the overlying
111 Victor Bay Formation by a pronounced subaerial unconformity (Turner, 2009, 2011). The Victor
112 Bay Formation represents rapid marine inundation of this irregular landscape and development
113 of a southwest-deepening mixed carbonate ramp. The Victor Bay basin was then tilted to
114 produce uplift in the west and drowning in the east, recording basin inversion during poorly
115 understood contractional deformation (Sherman et al., 2002). The upper Bylot Supergroup
116 consists predominantly of terrigenous clastic deposits of the Nunatsiaq Group, which has been
117 interpreted to represent an influx of sediment onto a complex topography associated with uplift
118 to the distant west of the basin (Knight and Jackson 1994; Sherman et al., 2002).

119 In summary, the tectonostratigraphic history of the Bylot Supergroup is complex, and
120 includes subtle regional subsidence during deposition of the Nauyat and Adams Sound
121 formations, enhanced subsidence that formed grabens and controlled facies distribution during
122 deposition of the Arctic Bay, Iqqittuq, Angmaat, Ikpiarjuk and Nanisivik formations, and
123 pronounced differential uplift both after deposition of the Angmaat and Nanisivik formations,
124 and again after deposition of the Victor Bay Formation. The Nunatsiaq Group (Knight and
125 Jackson, 1994) is of uncertain tectonostratigraphic affinity.

126 Mafic dykes of the Franklin swarm (ca. 723-713 Ma; Heaman et al., 1992; Pehrsson and
127 Buchan, 1999; Denyszyn et al., 2009) intruded along pre-existing graben-related fracture systems
128 in the Neoproterozoic (Figs. 1,2A). Most gabbro-hosting structures show little to no evidence of
129 fault displacement; they were deep-seated regional fracture systems inherited from the early
130 graben, and experienced episodic reactivation and fracture propagation into overlying units.

131 Proterozoic rocks are unconformably overlain by undeformed Paleozoic carbonate and
132 terrigenous clastic strata.

133 3. METHODS

134 Three stratigraphic sections, each hundreds of metres thick, were measured and described
135 in detail in the deeper-water, western area of the upper Arctic Bay Formation. At all three
136 locations, strata dip gently (6° to 18° northeast) and are well exposed in steep valley-walls.
137 Valley-side exposures were examined for evidence of features that record evidence of
138 syndepositional tectonic activity. Two of the sections were measured through the upper part of
139 the formation at Shale Valley, east of the end of Adams Sound (Figs. 2B,C and 3A,C; SVC and
140 SVE; 5 km apart). Samples of all rock types present were collected for laboratory analysis using
141 a rock hammer and pocket knife after excavation of the material to a depth of ~ 10 -30 cm. The
142 area was glacially eroded and exposed only $\sim 8,000$ years ago, so deep weathering profiles are not
143 present, but the material is inevitably to some extent weathered. A third section through the
144 entire Arctic Bay Formation was measured on the north side of the Alpha River valley, 10 km
145 west of Tremblay Sound and 90 km southeast of Shale Valley (Fig. 3B), but no true black shale
146 is exposed there (although a thick covered interval may contain black shale).

147 Rock samples from Shale Valley Central (SVC), were analysed for their major and trace
148 element compositions. Field samples were split into aliquots and dried at 70°C for 48 h to
149 eliminate moisture. Each ~ 10 g sample split was hand-milled in an agate mortar to a fine powder.
150 From this, a split was taken to perform a sequential 4-step loss-on-ignition (LOI) analysis on a
151 single aliquot. Volatiles were driven off at 105°C in pure N_2 to quantify any remaining moisture,
152 and then 3 steps of ignition under pure O_2 were undertaken, to successively burn off sulphides

153 (up to 371°C), reduced carbon (up to 500°C) and carbonate and structural OH (up to 1000°C). A
154 separate split was used to analyse hydrocarbon maturity and TOC by RockEval pyrolysis
155 (Geological Survey of Canada – Calgary). A further split was taken to determine major elements
156 by XRF (GeoLabs, MNDM), including total LOI. Finally, a 100 mg split was taken for trace
157 element analysis at Laurentian University. Samples were digested by HF-HNO₃ at 170°C in
158 Teflon beakers on a hotplate for 72 hours. This method does not completely digest zircon (e.g.,
159 Marx and Kamber, 2010) but has no adverse effect on transition metal concentrations, which are
160 the main focus of the present investigation. Due to the high reduced C content of the samples, the
161 fluoride residue that was obtained after digestion was twice attacked with 3mL of concentrated
162 HCl to help break down unwanted organo-metallic complexes, which might otherwise
163 compromise the metal yield. The chlorides were then twice converted with concentrated HNO₃
164 in preparation for dilution. The method was verified by analysis of the USGS shale standard
165 SCo-1, a silty calcareous marine shale. Data for this standard obtained at Laurentian University
166 are reported in Marx and Kamber (2010) and were found to be in good agreement with published
167 values. The solutions were analysed by ICP-MS using the method outlined in Kamber (2009).

168 Whole-rock Pb-isotope data were obtained for the 19 lowermost samples in the SVC
169 stratigraphic column. Lead was purified from the remaining stock solution of the trace element
170 digest according to the method of Kamber and Gladu (2009) and analysed with the method
171 described in Ulrich et al. (2010).

172 **4. LITHOSTRATIGRAPHY**

173 **4.1 Shale Valley**

174 Good exposures of the upper part of the Arctic Bay Formation are present on the
175 northeast side of Shale Valley. An unknown thickness of lower Arctic Bay Formation strata
176 separates these exposures from the underlying Adams Sound Formation. Intraformational
177 truncation surfaces (ITS) that record subtle slope failure healed by deposition of new sediment
178 over a low-angle fault scar are conspicuous on the hillsides and canyon walls of Shale Valley
179 (Fig. 3C); strata trace laterally to non-equivalent units, and changes in structural orientation are
180 evident across subtle contacts. Although no major intra-graben structural feature has been
181 mapped in the valley, it is probable that a major northwest-trending intra-graben fault is present
182 at or near Shale Valley (Fig. 1), based on the presence of ITSs and mound-related facies of the
183 Ikpiarjuk Formation,.

184 **4.1.1 Shale Valley Central (SVC)**

185 At SVC, the lowest exposed strata of the Arctic Bay Formation (unit 1; 31 m thick) consist of
186 shale-sandstone cycles 3 to 11 m thick, with hummocky cross-stratification (HCS) and gutter
187 casts (Fig. 2B,C). A blue-white, iridescent weathering sheen is present on some centimetre-scale
188 layers. Discontinuous rusty patches are rare. Medium-brown-weathering nodular lime mudstone
189 is present at 31 m. From 31 to 60 m (unit 2; 29 m thick), the succession is dominated by dark
190 grey shale with very minor sandstone interlayers. Medium-brown-weathering nodular lime
191 mudstone is present at 44 metres, and a carbonate intraclast debrite at 42 m. Black shale (unit 3;
192 176 m thick), some of it slightly calcareous, is non-cyclic from 60 to 88 m. From 88 to 236 m,
193 subtle shale-siltstone cycles, from 1 to 31 m thick, are marked by thin (generally <1 m), slightly
194 more resistant and slightly paler caps. Conspicuous, 60 cm thick, buff-weathering carbonate
195 layers with no distinctive structures are present at 122 and 136 m; a less pronounced carbonate
196 layer is present at 87 m. Rusty-weathering shale is conspicuous between 62 and 74 m. Unit 4 (43

197 m thick) consists of argillaceous carbonate and black shale. From 236 m to 260 m, the lime
198 mudstone is platy, resistant, and medium-grey-weathering, but from 260 to 279 m, lime
199 mudstone is black and recessive. At 256 m, a 2 metre-thick, lenticular slump fold and debrite are
200 present: contorted rafts of thin-bedded carbonate are entrained in the laterally discontinuous
201 mass, as are volumes of dolostone intraclast floatstone (debrite). Approximately 30-40 m of cliff-
202 forming, buff-weathering dolostone of the Ikpiarjuk Formation (unit 5) abruptly but conformably
203 overlie black shale at 279 m. The dolostone consists of matrix- to clast-supported, chaotically
204 disposed intraclasts: pale, equant dolostone clasts (Ikpiarjuk Formation) and tabular, medium
205 brown intraclasts (Nanisivik Formation; here geometrically equivalent to upper Ikpiarjuk Fm.).

206 **4.1.2 Shale Valley East (SVE)**

207 At SVE, the lowest unit (unit 1; 96 m thick) consists of shale-sandstone cycles with hummocky
208 cross-stratification, ripple cross-lamination and gutter casts; cycles are 3 to 24 m thick (Fig.
209 2B,C). A one m-thick, medium-brown-weathering nodular lime mudstone layer is present at 96
210 m. A blue-white, iridescent weathering sheen is locally present between 58 and 97 m. From 97 to
211 146 m (unit 2; 49 m), strata consist of dark-grey-weathering shale with scattered thin layers or
212 lenses that have a blueish weathering sheen. A second nodular limestone 40 cm thick is present
213 at 136 m. This unit is capped by a 1 metre-thick layer of cone-in-cone calcite. Black shale (unit
214 3; 67 m) dominates the section from 147 to 214 m. Subtle shale – siltstone cycles like those at
215 Shale Valley Central are 2 to 18 m thick. Scree material indicates that similar strata underlie a
216 covered interval from 214 to 237 m. Although most of the interval from 237 to 277 m (unit 4; 40
217 m) is covered, scree composition indicates that its composition is similar to that of the interval
218 between 277 and 303 m: black shale intervals (1-7 m thick) alternate with more resistant, one m-
219 thick bands consisting of clusters of layer-parallel, laminar concretions approximately 1 mm

220 thick and up to decimetres wide; these platy, discoidal, intrastratal concretions consist of dark,
221 coarse calcite crystals radiating unevenly from a central area. Rarely, this concretionary material
222 is present as concretion clasts in debrites, indicating that they formed early in the shallow
223 subsurface. Unit 5 is an in situ line of frost-heaved rubble of a 50 centimetre-thick layer of
224 tabular carbonate intraclast debrite belonging to the distal margin of an Ikpiarjuk Formation
225 mound. Unit 6 consists of shale scree ~1.5 m thick. Overlying, medium-brown-weathering,
226 homogeneous, medium-crystalline dolostone with a strong but ephemeral bituminous odour
227 when freshly broken belongs to the basal Nanisivik Formation (unit 7; in situ frost-heaved
228 rubble).

229 **4.2 Alpha River**

230 Strata of the Arctic Bay Formation are well exposed in canyons on the north side of the Alpha
231 River (Fig. 3B). The top of the Adams Sound Formation is present, although generally covered,
232 at the level of the river. Exposure ends at a plateau approximately 550 m higher, roughly where
233 Arctic Bay Formation shale passes to a complex array of overlying dolostone facies of the
234 Iqqittuq, Angmaat, Nanisivik and Ikpiarjuk formations (Figs. 2B, 3B).

235 The contact with the underlying Adams Sound Formation is not exposed, and a covered
236 interval between its inferred position and the lowest strata exposed is estimated to be 40-60 m
237 thick. Lithostratigraphic unit 1 consists of siltstone – sandstone cycles and is 332 m thick. In the
238 lowest Arctic Bay Formation strata (0-33 m), quartz arenite units 3-4 m thick are separated by
239 covered intervals. Overlying strata (33 m to 332 m) consist of shale-sandstone cycles. Cycles are
240 conspicuous from their weathering profiles, are 5 to 25 m thick, and thicken upwards, with a
241 decreasing sandstone component. Ripple cross-lamination, HCS and synaeresis cracks are

242 common to approximately 200 m, above which level sedimentary structures become fewer and
243 less distinct. Scattered cone-in-cone calcite concretions are present between 80 and 180 m. Unit
244 2 (181 m thick; 332 to 513 m) consists of shale-sandstone cycles that are 25 – 50 m thick. Thin
245 (<1 m), laterally continuous, medium-brown-weathering nodular limestone layers are present at
246 332 m and 406 m, and scattered carbonate nodules are present at 405-425 m and 460-480 m. A
247 resistant-weathering interval containing sandstone layers is conspicuous between 485 and 513 m
248 and contains hummocky cross-stratification and clasts of cross-laminated sandstone in sandy
249 debrites. The top of this interval contains one thin bed with polygonal cracks of unknown
250 derivation, and mudstone clasts in graded beds. Shale throughout the exposed section is various
251 shades of dark grey, and never truly black. It is commonly rusty-weathering near cycle bases
252 between 80 and 210 m, and two very conspicuous rusty-weathering intervals are present at 405-
253 415 m and 425-435 m; in both cases these rusty intervals are associated with nodular limestone.
254 An iridescent blue-white weathering sheen is present on some centimetre-scale layers between
255 33 and 418 m, and is especially well-developed between 317 and 333 m.

256 An estimated 125 m (unit 3a) that is covered separates the lower and upper exposures of
257 the Arctic Bay Formation. Shale-dominated strata of the upper Arctic Bay Formation (unit 3b;
258 148 m) are exposed on a slope between the first plateau above the Alpha River and the upper
259 plateau (regional high point). West of the measured section, this shale exposure is
260 stratigraphically equivalent to a level slightly below the base of a mound of the Ikpiarjuk
261 Formation (Bellevue Mound; Turner 2009). Above the grey shale succession is dolostone
262 exhibiting interlayered Nanisivik Formation and Angmaat Formation lithofacies. The grey shale
263 interval is laterally equivalent to an exposure of distal Iqqittuq Formation several hundred metres
264 to the east, which consists of interlayered shale, locally slump-folded ribbon to parted limestone

265 (thin-bedded lime mudstone with shaly interbeds or partings) and intraclast debrites (Fig. 3B).
266 Uppermost Iqqittuq Formation strata are affected by ITs in nearby exposures (Fig. 3D). Iqqittuq
267 strata are overlain by Angmaat Formation massive, pale, cherty dolostone. Given the geometric
268 relations among these units, tens of metres of uppermost Arctic Bay Formation are covered
269 between the uppermost exposed shale in the measured section and the mound-margin wedge, and
270 that the distal Iqqittuq Formation is equivalent to the uppermost Arctic Bay Formation. Owing to
271 scree cover, it was not possible to make an accurate geometric connection between interlayered
272 shale and limestone of the Iqqittuq Formation and the upper part of the Arctic Bay Formation.
273 Neither was it possible to sample the recessive, shaly interlayers of the outer Iqqittuq Formation.

274 The entire exposure above the covered interval (148 m; unit 3b) consists of dark-grey-
275 weathering shale with very rare, 10 cm-thick orange-brown dolomudstone layers.

276 No stratigraphic section was measured between Shale Valley and lower Alpha River, but
277 the Arctic Bay Formation has been examined immediately below its contact with the Nanisivik
278 Formation at numerous locations across the central Borden Peninsula. Almost invariably, the
279 uppermost Arctic Bay Formation in this region consists of black shale with intrastratal,
280 millimetre-thick, discoidal black concretionary layers of coarse calcite crystals like those in the
281 uppermost part of the SVE section.

282 **5. GEOCHEMICAL RESULTS**

283 **5.1 Geochemical data**

284 Analytical data for section SVC can be used to provide information about both sediment
285 provenance, which can be used to test depositional models based on field and sedimentological
286 data, and the chemical evolution of the water column below which the sediment accumulated.

287 The latter information is relevant to understanding Precambrian ocean-atmosphere evolution and
288 SEDEX potential of the basin.

289 **5.1.1 Rare earth element systematics**

290 The samples selected for laboratory analysis comprise a variety of lithologies:
291 conspicuous iridescent siltstone nodules, rusty sandstone, calcareous shale, black shale, rusty
292 shale, carbonate nodules, carbonate laminite and a carbonate slump. Although from a sediment
293 provenance perspective the siltstone nodules and carbonate rocks are of limited use, the upper
294 continental crust (here Mud of Queensland (MUQ); Kamber et al., 2005) normalised rare earth
295 element (REE) patterns of these lithologies are first briefly discussed (Fig. 4).

296 Two iridescent Mn-rich siltstone nodules (Table 1) have inverted V-shaped REE patterns
297 (into which Y is inserted at its position of relative ionic radius). The two important aspects of
298 these patterns (grey bars in Fig. 4A) are the absence of a Ce anomaly and the dip in Y compared
299 to neighbouring Dy and Ho. These features indicate that the Mn-rich nodules are probably
300 diagenetic in origin. For comparison, a representative REE pattern from a diagenetic Central
301 Pacific modern deep-sea Mn nodule (Fig. 4A) is shown. Although it is flatter overall (probably
302 due to different local lithology) it highlights the characteristic dip in Y and the lack of a Ce
303 anomaly. In contrast, hydrogenous Mn concretions have positive Y anomalies and very strong
304 positive Ce anomalies (e.g., Ohta et al., 1999). In a stratified ocean basin with a reduced lower
305 water mass, such as probably existed in the Mesoproterozoic, excess Ce may have been released
306 from Mn-oxyhydroxides, and the absence of a positive Ce anomaly, on its own, might not be
307 sufficient to distinguish between hydrogenous and diagenetic nodules. The combination with the
308 concentrations (1.05 and 4.33 wt%) of these two samples also support diagenetic uptake of the

309 REE by phosphates. However, the experimental data of Byrne et al. (1996) are at odds with
310 empirical observations regarding the behaviours of Y and Ho.

311 Sample SVC-60, a carbonate nodule, is also Mn-rich (13,800 ppm), and its REE pattern
312 has similarities with those of the siltstone nodules. The (weak) positive Ce anomaly of these
313 three samples is positively correlated with the total REE content ($r^2=0.810$, not shown) and with
314 the Co/(Cu+Ni) ratio ($r^2=0.829$, not shown). This pattern is similar to the systematics of modern
315 Mn nodules (e.g., Fig. 5a of Ohta et al., 1999) and indicates that the siltstone nodules are
316 diagenetic in origin, whereas the Mn-oxides in the carbonate nodule have a mixed hydrogenous
317 and diagenetic origin.

318 Moving up in the stratigraphy, regardless of stratigraphic height and carbonate content,
319 the remaining samples have similar patterns to upper continental crust, and appear flat when
320 normalised to upper continental crust. In more detail, when compared to the normalising upper
321 crustal composite, they show a 1.7-fold light over middle REE (LREE and MREE, respectively)
322 enrichment and flat MREE to heavy REE (HREE) patterns. The average normalised La/Gd of all
323 of the shales is 1.68 ± 0.29 , whereas the normalised Gd/Lu ratio is 1.17 ± 0.21 . This suggests that
324 the sediment was derived from a source that was more chemically evolved than average upper
325 continental crust.

326 The three non-nodular black shale samples richest in carbonate (SVC-71, 89 and 92) all
327 have negatively sloped normalised REE patterns (Fig. 4B), variably negative Ce anomalies, and
328 well-developed positive Y anomalies. These are features found in sediment that precipitated
329 from oxygenated water (required to explain the negative Ce anomaly) that is restricted from the
330 open ocean (explaining the negative overall slope; Kamber et al., 2004). Figure 4C shows the

331 REE pattern of the most calcareous shale (SVC-80), which shares the negative Ce anomaly and
332 the positive Y anomaly of the carbonate-rich samples (shown for comparison). These anomalies
333 are absent in shale samples that contain much less carbonate (shown as an example is SVC-86).
334 Carbonate admixture thus introduces Y and Ce anomalies into otherwise smooth REE patterns.

335 The REE systematics of the black shale, rusty shale and rusty sandstone exhibit an
336 interesting stratigraphic relationship (Fig. 5). There is a marked general decrease in overall REE
337 abundance from the bottom to the top of the formation. The overall REE content is very strongly
338 related to Al content ($r^2 = 0.983$), but only for samples with Al_2O_3 up to 11.5 wt%, indicating
339 that REE systematics in these samples is related to argillaceous content. However, for samples
340 with Al_2O_3 of >11.5 wt% (i.e., all of the shale samples), the correlation with total REE content
341 breaks down completely ($r^2 = 0.001$). Hence, the REE systematics of the shales are not
342 determined by the amount of clay.

343 The trend in REE content is accompanied by a change in REE pattern. The samples from
344 units 1 and 2 have essentially identically shaped REE patterns, pointing to a common sediment
345 source. The two lowermost analysed samples from unit 3 (66-84 m in Fig. 5) share the LREE
346 and MREE characteristics with the lower units, but are much more depleted in the heavy REE
347 (HREE), which signifies a change in sediment source. The trend continues in the samples
348 representing stratigraphic height between 94 and 109 m. The overlying, major part of unit 3,
349 represented by stratigraphic height from 132 to 231 m, have nearly constant REE patterns,
350 indicating a constant sediment source. The uppermost samples from unit 4 are the least enriched
351 in REE overall, which may indicate yet another slight change in sediment provenance.

352

353 **5.1.2 Quantification of effect of carbonate and clay content on REE systematics**

354 To test the qualitative finding that the REE patterns of the black shale were influenced by
355 carbonate, major element and 4-step LOI data were obtained (Table 1). Comparison of TOC data
356 with the LOI step from 375 to 500°C shows excellent correlation (although biased towards 15%
357 higher values in the latter, possibly due to loss of organic S and/or N). It is reasonable to assume
358 that combustion of carbonate occurs mainly between 500 and 1000°C under an O₂ atmosphere.
359 Observed values for this LOI step were compared with theoretical, calculated estimates by
360 attributing CaO and MgO to carbonate. When the entire Ca and Mg inventory is allocated to
361 carbonate, this comparison (Fig. 6A) shows that the observed and calculated LOI values
362 correlate very closely ($r^2 = 0.970$) with a slope of 0.96. This suggests that the original clay
363 minerals were low in Ca and Mg and had experienced a significant degree of weathering. It also
364 shows that even in the shale samples that have little Ca and Mg, a small amount of carbonate is
365 present. Furthermore, the stoichiometric Ca:Mg ratio shows that in all of the shales, the
366 carbonate is dolomite. Finally, the dolomitic composition is in strong contrast to that of the
367 crystal concretions (sample 09-SVC-89), which are pure Ca-carbonate. The strong influence of
368 the admixed dolomite on the REE pattern can be quantified when plotting (CaO+MgO) vs. the
369 Y/Ho ratio (Fig. 6B). The calcareous shale samples with high LOI have higher, super-chondritic
370 Y/Ho. Importantly, the calcite crystal concretion plots off the strong trend ($r^2 = 0.886$) defined by
371 the shales, which is caused by admixture of dolomite rather than of calcite.

372 In fine-grained siliciclastic sediment, clay content is commonly the dominant control over
373 the total lanthanide concentration (Σ REE). In the studied samples, however, Al₂O₃ controls REE
374 content only in the samples with low clay content (Fig. 6C). In the more clay-rich shales, the
375 correlation breaks down, suggesting that the REE content of the clay was variable, and as shown

376 in Figure 5, changed systematically with stratigraphic height. This is strong evidence for multiple
377 sources of clay-rich sediment.

378 The Al_2O_3 content of shale samples is strongly inversely correlated ($r^2 = 0.906$) with
379 $(\text{MgO}+\text{CaO})$, and so the combined evidence suggests that the shale is a mixture of two
380 sedimentary components: organic-C-rich clay, and dolomite. In those samples poorest in clay,
381 total lanthanide concentration is lowest, and therefore the marine REE characteristics are best
382 expressed (i.e., highest Y/Ho and most negative Ce anomaly).

383 **5.1.3 Conservative elements**

384 Ratios of conservatively behaving high-field-strength elements such as Zr/Hf, Nb/Ta,
385 Nb/Th show little change through the entire stratigraphic section. For example, there is only a
386 slight decrease in Nb/Ta with increase in stratigraphic height. The lowermost two samples (from
387 units 1 and 2) have average Nb/Ta of 15.03 ± 0.73 , followed by the unit 3 samples ($n=13$)
388 averaging 14.35 ± 0.42 , which are capped by unit 4 samples with an average of 14.03 ± 0.14 .

389 A general decrease in conservative, incompatible trace element concentration with time,
390 which is evident from the position of the REE patterns (Fig. 5), is conspicuous in plots of
391 concentration vs. stratigraphic height (Fig. 7A). Most of the commonly used immobile trace
392 elements, such as Ti, Zr, Nb, Hf, Ta, Th and the REE, as well as less commonly studied elements
393 such as W and Be, show marked decreases in concentration, on the order of 3- to 5-fold, from
394 bottom to top (Fig. 7A). This trend is not present in the compatible elements (e.g., Cr, Sc), the
395 most lithophile elements (e.g., Li, Cs), or in Al_2O_3 . There is no correlation between high-field-
396 strength element content and Al_2O_3 , indicating that clay minerals did not control supply of these
397 elements. Instead, the trend in high-field-strength element concentration (Fig. 7A) is likely

398 caused by a gradual decline of heavy mineral content. In the samples above the iridescent Mn-
399 rich siltstone nodules, the content of Ti, which is commonly used as a proxy for the heavy
400 minerals titanite, ilmenite and rutile, correlates positively with Th ($r^2 = 0.808$), Zr ($r^2 = 0.933$),
401 Nb ($r^2 = 0.913$) and Sn ($r^2 = 0.873$), suggesting that zircon and cassiterite were also important
402 detrital components.

403 **5.1.4 Metal chemostratigraphy**

404 The most interesting geochemical feature from both basin evolution and SEDEX
405 potential perspectives is that the general decline in incompatible conservative element
406 concentrations with stratigraphic height is not at all mirrored by many transition metals and
407 redox-sensitive elements. Instead, many of these metals show the opposite stratigraphic
408 evolution, becoming more enriched with increasing stratigraphic height, and reaching peak
409 values at ~200 m (Fig. 7B). The strongest variation in concentration is found for the elements
410 Mo (240x), Ag (33x), Cd (20x) and V (7.6x); Ni, Cu, Zn and U also show considerable
411 concentration increases that contrast with the decreasing trend of the high-field-strength
412 elements. The concentration increase is not monotonic with stratigraphic height, but shows a
413 systematic pattern with maxima between 145 and 200 m (Fig. 7B). Because the Nb/Ta and REE
414 systematics of the studied shales are different from average upper continental crust, it is not
415 feasible to calculate conventional enrichment factors. Instead, the detrital contribution to the
416 shale's metal inventory was removed by using metal ratios relative to conservative elements. The
417 resulting stratigraphic enrichment trends are smoother (Fig. 7C) and the overall concentration
418 contrast increases. For example, although the V concentration alone varies by a factor of 7.6, the
419 V/Nb ratio changes by a factor of 25.3. Similarly, Cd, whose elemental concentration range is
420 20, expresses a relative enrichment range (calculated as Cd/Th) of 77. The relative enrichment of

421 V and Cd peaks at 200 m (Fig. 7C), whereas Mo, U, Ni, Zn and Cu show two maxima, at ~145
422 m and ~200 m.

423 As is anticipated from the coherent behaviour in the context of stratigraphy, the
424 metal/conservative element ratios are strongly correlated. For example, V/Nb correlates with
425 Cd/Th ($r^2=0.846$) and Mo/Hf ($r^2=0.868$; Fig. 8A), which itself correlates, although not as strongly,
426 with U/Th ($r^2 = 0.339$), which in turn correlates with Ni/Cr ($r^2=0.535$; Fig. 6B). The possible origin
427 of these trends is discussed below.

428 **5.1.5 Pb-isotope systematics**

429 The whole-rock Pb-isotope data form a co-linear array in common Pb space (Fig. 9).
430 When the two most deviating samples (SVC-71 and 75) are omitted, the remaining 17 data
431 points define an isochron (MSWD = 1.6) of 1137 ± 98 Ma. Furthermore, 17 of the 19 data points
432 (excluding SVC-72 and 80) also yield an $^{238}\text{U}/^{206}\text{Pb}$ errorchron of 1067 ± 100 Ma, and 14 samples
433 (excluding SVC-51, 58, 72, 79 and 80) define an $^{232}\text{Th}/^{208}\text{Pb}$ errorchron of 1059 ± 120 . These are
434 three independent age constraints yielding a weighted mean of 1092 ± 59 Ma. This date, although
435 not very precise, is interpreted as the age of deposition and is in agreement with other
436 chronological constraints. In terms of original Pb-isotope composition of the sediment, the
437 isochron projects between the coeval mantle evolution composition and that of juvenile upper
438 continental crust. This suggests a mixed sediment supply dominated by juvenile sources, which
439 is in agreement with the high Nb/Ta ratio, other provenance indicators, and the local geology.

440

441 **6. INTERPRETATION**

442 **6.1 Basin Evolution and Provenance Information**

443 The evolution of the northwestern MIG during deposition of the Arctic Bay Formation
444 can be reconstructed using lithostratigraphic profiles from Shale Valley and Alpha River (Fig.
445 10) together with regional stratigraphy (Turner, 2009). A rift-like phase of the MIG initially
446 developed when geographically limited excessive subsidence formed a west-deepening ramp
447 (Arctic Bay Formation) flanked at its southern and northern margins by coarse terrigenous
448 detritus (Fabricius Fiord Formation; Jackson and Iannelli, 1981; Scott and deKemp, 1998) that
449 was deposited along the flanks of horsts veneered with Adams Sound Formation quartz arenite
450 and Nauyat Formation basalt. The northwestern, lower Arctic Bay Formation was initially
451 characterised by very gentle northwest-ward deepening in a regionally storm-dominated,
452 strongly cyclic, sandy to muddy setting. Waning of terrigenous input produced a protracted
453 upward-fining succession that was punctuated by deposition of rare benthic carbonate mudstone,
454 and eventually led to black shale deposition in the outer, northwestern parts of the muddy ramp.
455 In the southeastern MIG, carbonate deposition of the Iqqittuq Formation, laterally equivalent to
456 the black shale interval in the northwest, was punctuated by intervals of bedded sulphate-facies
457 evaporite.

458 Direct evidence of synsedimentary faulting during black shale deposition is present in the
459 form of ITSs, which represent areas where deep-water slopes failed and the resulting surfaces,
460 cutting at low angles across underlying strata, were overlain by new sediment layers parallel to
461 scar surfaces. Such ITSs and the associated evidence for locally variable sediment composition
462 and accumulation rates are present both at Shale Valley (lower and upper Arctic Bay Formation)
463 and at Alpha River (distal Iqqittuq Formation, which is laterally equivalent to upper Arctic Bay
464 Formation black shale). Given the very subtle slope gradient from the central part of the MIG to

465 the northwestern part (at most several tens of metres of topographic differential, across a lateral
466 distance of ~100 km), this type of over-steepening and gravitational failure of slopes cannot have
467 been caused by normal sedimentary or subsidence processes, and was instead probably a product
468 of tectonic adjustment along major intra-graben faults. The ITSs documented to date are in the
469 immediate vicinity of such faults. In some cases the failure direction was at a high angle to or
470 even opposed to the regional northwest-deepening gradient of the basin floor, supporting the
471 interpretation that the ITSs were caused by local over-steepening associated with intermittently
472 active fault systems.

473 At Shale Valley, an ITS that affected SVE but not SVC removed part of the black shale
474 of unit 3. Consequently, unit 3 black shale is markedly thinner at SVE than at SVC, owing in
475 part to removal of at least several tens of metres of section at a conspicuous ITS. Unit 4 is
476 strikingly different in the two Shale Valley sections: the SVC location received resedimented,
477 slumped carbonates, inferred to be related to a mound- and fault-related slope, whereas the SVE
478 location accumulated black shale in which concretionary calcite grew in the shallow subsurface.
479 Pronounced differences in the thicknesses of these geometrically (but not temporally or
480 paleoenvironmentally) equivalent units are supported by geochemical evidence (described
481 below).

482 Indirect evidence of subtle syndepositional faulting is provided by large, elongate, deep-
483 water (sub-photic zone) carbonate mounds of the Iqqittuq Formation, which grew during
484 deposition of the upper Arctic Bay Formation black shale, only in the vicinity of major intra-
485 graben faults (Turner, 2009). It is inferred that fissures related to the faults reached the sea-floor
486 and dilated in response to subtle tectonism to serve as conduits for fluids associated with mound
487 development.

488 Shale deposition ended abruptly owing to tectonic adjustment that produced a semi-
489 restricted, terrigenous-sediment-free basin differentiated by a tectonic barrier into two zones. In
490 the northwestern zone, carbonate precipitated either in the water column or directly on the sea
491 floor, and accumulated as monotonous, millimetrically laminated deep-water dolostone,
492 (Nanisivik Formation). The southeastern zone was characterised by cyclic accumulation of
493 lagoonal to peritidal carbonate rocks with abundant sea-floor carbonate dendrites and
494 “abiogenic” stromatolites, which are widely recognised as indicators of evaporative carbonate
495 oversaturation (Angmaat Formation; Turner, 2009).

496 Prior to development of the graben, the paleoclimate had been humid enough for the
497 existence of river systems that supplied quartzose sediment to the predominantly marine Adams
498 Sound Formation. By the time that the prograding, carbonate-dominated Iqqittuq Formation was
499 accumulating in the southeastern MIG (laterally equivalent to black shale of the upper Arctic
500 Bay Formation), the climate had become arid enough to result in deposition of bedded sulphate-
501 facies evaporites (Jackson and Cumming, 1981; Kah et al., 2001). Still later, the climate became
502 humid once again, as indicated by karstification of the upper Victor Bay Formation (Sherman et
503 al., 2002).

504 The inferred low Ca and Mg content of the clay in the analysed shale supports a strongly
505 weathered sediment source. In terms of provenance, the shale has several notable features that
506 agree with respect to sediment provenance. The high Nb/Ta (14-15) of the sediment is much
507 higher than typical continental crust (~11.5; Kamber et al., 2005) and is a strong indication that
508 the sediment source was dominated by basalt (with Nb/Ta>15) with a only a subordinate felsic
509 component. This finding is supported by the REE patterns of the shales, which show greater
510 LREE than MREE and only mild MREE over HREE enrichment. Such kinked REE patterns

511 (e.g., Xu et al., 2005) as well as unusually high Nb/Ta (Murphy et al., 2003) are characteristic of
512 alkali basalts associated with continental rift magmatism.

513 **6.2 Origin of metal enrichment**

514 The geochemical data (Figs. 7B,C, 8) document consistent enrichment of a number of
515 metals in the black shale part of the SV sections, chiefly in unit 3. In order to assess the extent to
516 which this enrichment could be relevant to the SEDEX potential of the basin, it is necessary to
517 determine the cause of metal enrichment. The first-order question is to determine whether the
518 metal enrichment occurred during sediment settling (by scavenging from the water column or
519 along the water/sediment interface) or during diagenesis.

520 The most widely discussed mechanism (e.g., Lyons et al., 2009) for redox-sensitive metal
521 enrichment in black shale involves a chemically stratified water column and a particle shuttle
522 (organic, metal oxyhydroxides, or sediment) that acts as a direct carrier of metals or helps to
523 maintain physico-chemical conditions that are conducive to metal incorporation into the
524 sediment (e.g., euxinia above or within the sediment). The presence of anoxic bottom-water
525 conditions during deposition of black shale of SVC units 3-4 is not in question. The TOC, as
526 determined by RockEval pyrolysis as well as the third step of LOI (375 to 500°C), which is
527 dominated by reduced C, averages between 5 and 6 wt%, suggesting that high primary
528 productivity in the upper part of the water column may have helped to maintain anoxic and, at
529 times, euxinic conditions at depth, in a rift basin that may have been restricted and may have had
530 a primary chemocline typical of the Mesoproterozoic ocean as a whole (e.g., Canfield et al.,
531 2008). Most samples have high Fe/Al ratios (on average 0.71, but up to 2.1) that are very similar
532 to other Precambrian shale deposits (e.g., Raiswell et al., 2011). Although no Fe-speciation and

533 quantitative S data were obtained, the combination of high TOC, high Fe/Al and S contents, and
534 qualitative Fe/S ratios similar to pyrite stoichiometry, strongly suggests that euxinic conditions
535 were reached at times during deposition of the shales of unit 3.

536 The negative Ce anomalies in limestone beds and nodules argue for an uppermost water
537 column that was sufficiently oxygenated to convert Ce^{3+} to Ce^{4+} . Conversely, the Mn- and
538 phosphate-rich samples have slight positive Ce anomalies, suggesting that at depth, below the
539 chemocline, Ce may have been reduced back to Ce^{3+} . Furthermore, the limestone nodules have
540 positive Y/Ho, whereas the phosphate-rich Mn nodules have subchondritic Y/Ho. These
541 systematics afford an opportunity to propose an origin for the dolomite, which may have
542 precipitated from the uppermost, oxygenated water column and rained down as particles, or may
543 have formed within the sediment due to alkalinity changes driven by sulphate-reducing bacteria
544 [i.e., the “organogenic dolomite” described by Mazzullo (2000)]. The REE systematics argue in
545 favour of a upper-water-column origin for the carbonate in the black shales, but allow for the
546 possibility of diagenetic conversion of limestone to dolomite, because this process does not
547 necessarily change REE patterns (e.g., Webb et al., 2009). Combined C- and O-isotope evidence
548 would be needed to further elucidate the origin of dolomite.

549 Despite these coherent observations in favour of a redox-stratified basin with anoxic
550 bottom waters, there is no straightforward correlation between TOC, Fe/Al ratio, Mn, and
551 qualitative S content.

552 None of the samples other than the Mn-rich nodules, contains an appreciable positive Ce
553 anomaly (Figs. 4 and 5). In a redox-stratified basin, Ce is either oxidised in the shallow water
554 column, or once trivalent Ce is adsorbed onto the hydroxide ion (Bau and Koschinsky, 2009),

555 and is then transported with particles into the anoxic, deeper water, where it is released. For this
556 reason, both Fe and Mn-hydroxides are very strongly enriched in Ce relative to neighbouring La
557 and Pr, giving rise to very characteristic REE patterns with strong positive Ce anomalies (e.g.,
558 Ohta et al., 1999). Such an anomaly is not present in SVC shale. Manganese oxides also have
559 sub-chondritic Y/Ho ratios, another feature that is absent in the shale. From the combined
560 evidence, it can be concluded that the total authigenic metal budget in the sediment has multiple
561 origins that include clastic sedimentation rate, reduced C supply, Mn- or Fe-oxyhydroxide
562 shuttles, availability of HS^- , and other factors, none of which exerted dominant control. The lack
563 of correlation among these chemical proxies is not unique to the black shale studied here (e.g.,
564 Raiswell et al., 2011), and does not automatically imply post-depositional modification (e.g.,
565 Cruse and Lyons, 2004).

566 Molybdenum enrichment reaches concentrations of 50 ppm (Fig. 7C; the highest reported
567 values for Mesoproterozoic black shales; see Scott et al., 2008) and offers an opportunity to
568 explore possible reasons for overall metal enrichment. Elevated Mo concentrations generally
569 result when Mo that is complexed with organic C encounters elevated concentrations of HS^-
570 either in the water column or in sediment pore waters, at which stage it is scavenged effectively
571 as thiomolybdate (e.g., Algeo and Tribovillard, 2009). In the Arctic Bay Formation, despite the
572 very strong Mo-enrichment in many of the studied samples, there is no discernible correlation of
573 Mo/Hf with Fe/Al ($r^2 = 0.207$), estimated reduced C content ($\text{LOI}_{375-500}$; $r^2 = 0.004$) or qualitative
574 S content ($r^2 = 0.017$). Lack of correlation persists regardless of whether clastic input is corrected
575 for with Al, Ti or as enrichment factors. Instead, there is a significant positive correlation with
576 the Y/Ho ratio ($r^2 = 0.650$) and with the carbonate (dolomite) content ($\text{LOI}_{500-1000}$; $r^2 = 0.659$). In
577 view of the strong correlation of Mo enrichment with other metals such as V (Fig. 8A,B), it is

578 not surprising that V/Nb is also strongly correlated with Y/Ho ($r^2 = 0.815$, Fig. 8C). Dolomite
579 content of the black shale correlates with metal enrichment and so the most parsimonious
580 explanation for metal enrichment in the shale is the activity of a dolomite (or dolomite precursor)
581 particle shuttle or a dolomite metal trap in the sediment. No attempt was made to analyse the
582 fine-grained carbonate separately for metal content.

583 One observation from the present dataset, however, indicates that the stratigraphic metal
584 distribution as measured today may not be completely primary. Authigenic metal enrichment
585 should be strongly associated with elevated Co concentrations. For example, hydrogenous
586 FeMn-oxide crusts have very elevated Co/(Ni+Cu) (e.g., Calvert and Price, 1977). However, the
587 SVC black shale data show no correlation whatsoever between Co/(Ni+Cu) and Mo, V, C or U
588 enrichment. Other studies have previously documented complex redox-sensitive authigenic metal
589 distributions in black shale (e.g., Pennsylvanian black shale from Oklahoma in Cruse and Lyons,
590 2004). In actively rifting basins that experience low-T hydrothermal discharge, additional metal
591 sources other than a stratified water column may significantly increase metal content (as
592 proposed for Devonian black shale in Yukon; Hulbert et al., 1992), which might lead to the
593 dissociation of Co vs. (V,Mo, U) trends.

594 An alternative working hypothesis is that the metal distribution, as it is found today, was
595 affected by secondary basinal fluids (e.g., Leventhal, 1991). In this model, the overall
596 enrichment of redox-sensitive metals in the entire sediment body was originally caused by metal
597 transfer into the sediment in a stratified water body with anoxic and at times euxinic bottom
598 waters (as discussed above). Although the stratigraphic relative metal enrichment appears at first
599 glance (e.g., Fig. 7) to be systematic, there is a conspicuous lack of correlation between metal
600 enrichment and Fe/Al, Mn, or TOC content. Instead, the metal enrichment appears to track the

601 dolomite content of the shales. If the dolomite formed in situ due to increased alkalinity caused
602 by sulphate-reducing bacteria, for example, it is possible that the metals were enriched locally
603 from pore waters. Alternatively, if the dolomite represents a primary sediment component, it
604 could be hypothesised that when diagenetic fluids migrated through the sediment column upon
605 compaction, the carbonate in the calcareous shales would have acted as a redox trap, elevating
606 the fluid's pH, affecting metal solubility, and causing precipitation.

607 Regardless of the exact mechanisms that caused metal enrichment, an important
608 observation from this study is that absolute concentrations of redox-sensitive metals in black
609 shales (e.g., Mo) must be used with caution to infer ancient ocean conditions, particularly when
610 elevated metal contents are associated with subtle carbonate-bearing horizons.

611 **6.3 SEDEX potential of the black shale basin**

612 Depositional settings that are associated with SEDEX/CD potential commonly exhibit
613 evidence of synsedimentary extensional faulting under euxinic deep water, starvation of
614 terrigenous and precipitated sediment, and topographic compartmentalisation of the basin floor
615 into small-scale sub-basins whose bottom-water chemistry could evolve independently of other
616 parts of the basin (Leach et al., 2005; Goodfellow, 2007; Goodfellow and Lydon, 2007; Leach et
617 al., 2010). Also favourable is the presence of evaporites in strata that are laterally and temporally
618 related to the black shale interval, which create dense brines that sink and acquire metals as they
619 migrate through buried aquifer units (e.g., Russell et al., 1981; Goodfellow et al., 1993; Leach et
620 al., 2005). Virtually all of the structural and stratigraphic preconditions for SEDEX/CD
621 deposition are amply evident in the Arctic Bay Formation. These positive attributes are then
622 greatly reinforced by the basin's geochemical history.

623 Geochemical evidence depicts a lengthy time during which the deep-water area of the
624 upper Arctic Bay Formation was occupied by stratified water with reducing or euxinic bottom
625 water. The water column in the basin was therefore capable of converting metals supplied as
626 solutes into a particulate component of the sea-floor sediment. Of the base metals present in
627 anomalous concentrations throughout the black shale interval, Zn covaries with Mo, suggesting
628 that, like Mo, it was removed from solution by reduction. Unlike Mo, however, Zn, and to a
629 lesser extent Ni, were probably not sourced from terrestrial weathering, and so if the enrichment
630 in Zn and Ni are primary, these metals would have been supplied by vented fluids. Both the
631 carbonate rocks and the black shale lack positive Eu anomalies, which is evidence against high-T
632 (i.e., >250°C) hydrothermal-vent-type fluids (e.g., Kamber, 2010) and suggests that the fluids
633 were probably of the off-ridge-axis, diffuse, low-T variety. Zinc and Ni could have been supplied
634 from leaching of the underlying pile of fine terrigenous sediment by basinal brines. This is a
635 plausible source for the Zn, and commonly inferred in SEDEX systems (e.g., Russell et al., 1981;
636 Goodfellow et al., 1993). The stratigraphic spacing of samples analysed in this study was coarse,
637 and intervals with elevated metal concentrations could easily have been missed. Although Zn
638 concentrations do not reach economically meaningful concentrations at the stratigraphic levels
639 tested in this study at this particular location, the broad stratigraphic, structural and geochemical
640 conditions for a SEDEX system were present.

641 Mineralising systems other than SEDEX-type should be considered for the black shale
642 succession. Molybdenum and other redox-sensitive elements were stripped from sea water
643 during black shale deposition, and it could have been possible, with a very low sedimentation
644 rate of fine terrigenous material, for Mo, U, and other redox-sensitive metals to reach high
645 concentrations on the sea-floor. There is no evidence of such elevated concentrations in the

646 stratigraphic levels tested at the one location addressed by this study, but the sample spacing was
647 coarse, such that geochemically interesting intervals may have been overlooked by this survey.
648 The sedimentary and geochemical evolution of other black shale sub-basins of the Arctic Bay
649 Formation may have been more favourable for deposition of redox-sensitive metals.

650 Unit 1 (0-31 m) of the succession at SVC contains siltstone nodules to layers that are
651 conspicuous in outcrop from their iridescent, violet-white weathering sheen. These units are
652 characterised by elevated concentrations of Mn, Pb, As, Ni, Zn and Fe, a composition that
653 resembles that of modern deep-sea Mn nodules. If this composition was primary, the abundance
654 of Mn in these layers reflects deposition below a chemocline, but one characterised by a
655 geochemical composition that is markedly different from that of the later black shale basin, as
656 reflected by REE composition.

657 **6.4 Depositional age**

658 In the eastern MIG, the Arctic Bay Formation is overlain by carbonate-dominated strata
659 of the Iqqittuq and Angmaat Formations (collectively formerly known as the Society Cliffs
660 Formation), the Victor Bay Formation, and the Athole Point Formation. This carbonate-
661 dominated succession shows a distinctive $\delta^{13}\text{C}$ curve that has been correlated with carbonate
662 strata in Mauritania (Kah et al., 1999) and dated at 1105 ± 12 to 1109 ± 37 Ma (Re-Os; Rooney et
663 al, 2010). The identical (within its larger error) U-Th-Pb shale depositional age of 1092 ± 59 Ma
664 yielded by the present study for black shale that is temporally equivalent to part of the carbonate-
665 dominated succession (i.e., equivalent to upper Iqqittuq Fm.) fits well with this chronology. The
666 succession may be significantly younger than a previous estimate of approximately 1199 ± 24
667 Ma (Kah, unpublished data, in Kah et al. 2001).

668 **7. DISCUSSION**

669 **7.1. Dynamics of dissolved molybdenum in the Mesoproterozoic ocean**

670 Black shale geochemistry is rapidly becoming a major source of information used to infer
671 planetary atmospheric evolution and ocean ventilation (e.g., Scott et al., 2008; Dahl et al., 2011).
672 The present study provides a large new dataset that can be used to qualify some of the inferences
673 made from global compilations.

674 The REE patterns of the Arctic Bay Formation calcareous black shale suggest carbonate
675 (dolomite or dolomite precursor) formation in oxygenated water. Given the high TOC and
676 reduced metal content of the shales, this provides strong evidence for a rift basin in which a
677 chemocline had developed. It also supports a globally oxidizing atmosphere. However, the
678 observation of anoxia in a rift basin does not imply that the global deep ocean was anoxic.

679 A major argument in favour of an anoxic and sulphidic post-1.8 Ga Proterozoic deep
680 ocean is the relatively low Mo content of black shale, in comparison with Paleozoic equivalents
681 (Scott et al., 2008). The reasoning behind this argument is that the marine inventory of Mo in a
682 sulphidic ocean could never have increased greatly because of the continuous drawdown of Mo
683 into pyrite across the entire ocean floor. The maximum Mo concentrations encountered in this
684 study (~50 ppm) agree with global compilations (e.g., Scott et al., 2008), but the present dataset
685 is from a localised rift graben basin, rather than a setting fully linked to the global ocean. If the
686 basin water was not in full exchange with the open ocean, the relatively limited Mo enrichment
687 encountered in the black shale can only be used to infer that the Mo concentration in this
688 restricted basin never reached the levels of the Phanerozoic ocean. This observation does not
689 preclude the possibility that open seawater had a much higher Mo concentration.

690 Because the primary source of Mo to the ocean is generally assumed to be fluvial, the
691 number and distribution of rift basins themselves may also be relevant. Molybdenum supplied by
692 rivers to rift basins would be removed effectively on account of the likely anoxic/euxinic nature
693 of these basins. Hence, at times of supercontinent rifting, the global ocean Mo inventory may be
694 substantially lower than during times of dispersed continents because of the combined effects of
695 reduced fluvial flux and enhanced scavenging of metal into anoxic/euxinic rift basins.

696 Regardless of the issue of local water versus open seawater chemistry, a further
697 complication arises from the present study: the possibility of secondary enrichment of redox-
698 sensitive metals, such as Mo, into thin stratigraphic horizons. Geochemical studies generally
699 assume that Mo content is a reflection of primary sediment chemistry and that Mo is hosted by
700 pyrite (e.g., Lyons et al., 2009). At Shale Valley Central, however, high Mo content is associated
701 with high dolomite content and may not be a true reflection of the basin's water chemistry. This
702 may not be a unique situation. For example, the proposal of a brief Neoproterozoic global oxidation
703 event was based in part, on elevated Mo contents in the McRae shale (Hamersley Basin,
704 Australia; Anbar et al., 2007). The enriched interval in the 30 m-thick S1 shale stratigraphy
705 coincides exactly with the occurrence of a grey carbonate band (Fig. 1 of Anbar et al., 2007).
706 The findings from the Arctic Bay Formation suggest that secondary enrichment in Mo may be an
707 alternative explanation for the high Mo/Al ratios in the McRae shale. In the present case, it is
708 possible to test the origin of the metal enrichment because of the wide array of reported chemical
709 data. Many black shale chemistry datasets that are used to make global-scale inferences do not
710 report the full suite of metal concentrations, which means that the possible role of diagenetic
711 overprinting cannot be assessed.

712 **8. SUMMARY**

713 Rare-earth elements and redox-sensitive metals together with lithostratigraphic analysis
714 indicate that late Mesoproterozoic black shale of the Arctic Bay Formation (1092±59 Ma; Th-U-
715 Pb whole-rock) was deposited under a stratified oxidised-euxinic water mass in an actively
716 extensional basin. The highly redox-sensitive metals, such as Mo, V and U, were strongly
717 enriched into the black shale and show coherent behaviour. These elements were probably
718 supplied from the water column via a particle shuttle as well as possibly from diffuse venting.
719 Less redox-sensitive metals (e.g., Zn and Ni) were also enriched, but to a lesser extent, and may
720 not have been carried by a shuttle. The association of metal enrichment with dolomite content
721 indicates either that the dolomite (or its precursor) was the shuttle, or that secondary
722 redistribution of metal resulted in enrichment in the most dolomitic shale horizons. Although
723 there is no evidence for pronounced base-metal enrichment caused by SEDEX-type venting at
724 the one black shale location studied in detail, the strong evidence of a chemocline and the
725 establishment of bottom-water euxinia indicates that the geochemical conditions necessary for
726 metal precipitation were present. Black shale of the Arctic Bay Formation was deposited during
727 what appears to have been a global nadir of SEDEX/CD deposit formation, but its basin would
728 have been geochemically capable of precipitating and concentrating base metals had they been
729 supplied at a sufficient rate. The Shale Valley Central dataset provides insight into the
730 complexity of water and sediment chemistry evolution in a restricted basin but cannot be used to
731 infer global oceanic conditions.

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741

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894 FIGURE CAPTIONS

895 Fig. 1. (A) Two study locations (black stars) on a tectonic map of the Borden Basin, of which the
896 main trough is the Milne Inlet Graben. Only the three formations relevant to base-metal
897 prospectivity are shown (other parts of the Bylot Supergroup, as well as crystalline basement and
898 Paleozoic strata are shown in grey). (B) Location of Borden Basin among the other Bylot basins.
899 (C) Location of map (A) in Canada's Arctic islands.

900

901 Fig. 2. (A) Generalised stratigraphy of the Bylot Supergroup, after Jackson and Iannelli (1981),
902 Sherman et al. (2002), and Turner (2009), including tectonostratigraphic interpretation and
903 diagrammatic depiction of geometric complexity among Arctic Bay Formation (grey & black
904 fill) and associated carbonate formations. IQ = Iqqittuq Fm.; AT = Angmaat Fm.; IK = Ikpiarjuk
905 Fm.; NA = Nanisivik Fm.; FF = Fabricius Fiord Fm. The Fabricius Fiord Formation is present
906 only along the margins of the Milne Inlet Graben. (B) Stratigraphic panel of part of the Arctic
907 Bay Formation with proposed correlations for the two localities (Shale Valley (SV) and Alpha
908 River (AR)) addressed in this study. Vertical bars to right of stratigraphic column indicate
909 iridescent weathering sheen (grey) and rusty weathering (black) in field exposures. Note scale
910 difference between SV and AR. (C) Expanded view of two sections at Shale Valley (SV Central
911 and East), showing inferred synsedimentary truncation surface, lateral thickness and facies
912 changes, and relationship to the overlying flank of a carbonate mound. Four lithostratigraphic
913 units are identified in the two areas: (1) siltstone and sandstone of a northwest-deepening storm-
914 dominated ramp; (2) sub-storm-wave-base siltstone with rare but conspicuous carbonate nodules;
915 (3) black shale with subtle grain-size cycles; and (4) black shale with early diagenetic discoidal

916 calcite concretions, sooty carbonate layers, and debrites of clasts of these lithofacies. SVC and
917 SVE are separated by 5 km, but had distinctly different sedimentary histories because of sea-
918 floor topography associated with syndepositional activity of a nearby fault, as well as owing to
919 their different proximities to the tapering flank of an Ikpiarjuk Fm. carbonate mound in the
920 uppermost parts of the sections.

921

922 Fig. 3. Field exposures of Arctic Bay Formation and associated strata. (A) Shale Valley Central
923 (SVC) section in westernmost Milne Inlet Graben, with C. Gilbert in foreground. Section SVC
924 follows gully in centre of photo. The lower part of the Arctic Bay Formation is not exposed. (B)
925 Arctic Bay Fm. as exposed on northeast side of Alpha River Valley near Tremblay Sound.
926 Contact with underlying Adams Sound Fm. is just above river level. The geometric relationships
927 among Arctic Bay Formation and overlying carbonate units of the southeastern (Iqqittuq and
928 Angmaat fms.) and northwestern Milne Inlet Graben (Ikpiarjuk and Nanisivik fms.) are
929 displayed. The uppermost Arctic Bay Formation is laterally equivalent to the upper part of the
930 outermost Iqqittuq Fm. (carbonate ramp) and also to deep-water carbonate mounds of the
931 Ikpiarjuk Fm. Section AR follows red dashed line. Strata dip away from viewer at various
932 attitudes between 10° and 20°. Elevation difference between valley bottom and plateau is
933 approximately 550 m. (C) An intraformational truncation surface (ITS; healed low-angle slope-
934 failure scar) in the Arctic Bay Formation at on south wall of Shale Valley records
935 syndepositional tectonic activity: gravitational failure of a shale slope and healing of the scar by
936 continued shale deposition. The eastward down-cutting truncation direction is opposite to the
937 overall gentle northwestward deepening of the basin, supporting the idea of local slope
938 development associated with nearby syndepositional fault activity. (D) Subtle ITS in the

939 outermost ramp strata of the Iqqittuq Fm. on the south wall of Alpha River Valley, near where
940 Iqqittuq Fm. passes laterally to upper Arctic Bay Fm. Eastward down-cutting is antithetical to the
941 overall westward deepening indicated by lithostratigraphic pattern, attesting to the presence of a
942 local slope caused by movement on a rift-related intra-graben fault (the Tikirarjuaq Fault Zone is
943 in the valley toward the viewer). Exposure is approximately 500 m thick.

944

945 Fig. 4. Upper continental crust normalised (using MUQ of Kamber et al., 2005) REE patterns of
946 non-shale samples analysed by ICP-MS. Panel A shows two siltstones (SVC-51 and 55) and one
947 carbonate nodule (SVC-60) compared to representative sub-recent diagenetic central Pacific Mn
948 nodules (average of samples HW-2 and HW-15 from Ohta et al., 1999). Panel B shows non-
949 nodular carbonate samples. Panel C compares the most calcareous shale sample (SVC-80),
950 average carbonate (from panel B) and a carbonate-free shale (SVC-86). For explanation see text.

951

952 Fig. 5. Upper continental crust normalised (using MUQ of Kamber et al., 2005) REE patterns of
953 shale samples as function of stratigraphic height. Samples were grouped into 7 height intervals.
954 Relative position is indicated by weight of shading. For explanation see text.

955

956 Fig. 6. (A) Correlation between measured 500-1000°C LOI and that calculated when attributing
957 all Ca and Mg in sample to carbonate (excepting the two Mn-rich siltstone nodules). (B)
958 Correlation between carbonate content (approximated with CaO + MgO) and Y/Ho ratio. Note
959 that calcite crystal concretion is shown as open circle and plots off the trend defined by the

960 shales. (C) A positive trend between Al_2O_3 content and sum of lanthanide concentration (note
961 that this does not include Y) is found up to Al_2O_3 of 11.5 wt%, whereas in the more aluminous
962 samples, REE content is not at all related to argillaceous content.

963

964 Fig. 7. Chemostratigraphic plots of samples, omitting siltstone and carbonate nodules. (A)
965 Coherent stratigraphic patterns for Zr, Nb and Tm. (B) Stratigraphic trends defined by redox-
966 sensitive elements U, Mo and Ni. (C) Plots of ratios of redox-sensitive metal relative to
967 immobile lithophile element shows steady, strong enrichment in Mo, Cd and V from 50 m to a
968 peak at 200 m before returning to low values above 250 m.

969

970 Fig. 8. Binary co-variation diagrams. (A) Strong positive correlation between relative V and Mo
971 enrichment (see also Fig. 9C). (B) Moderately strong correlation is also found between relative U
972 and Ni enrichments. (C) Very strong positive correlation between relative V enrichment and
973 Y/Ho ratio. See explanation in text.

974

975 Fig. 9. Common Pb-isotope diagram showing 17 whole-rock data-points used for isochron
976 calculation (solid small diamond symbols with 1-sigma error bars). Note two outliers (solid
977 squares with error bars). For reference, the evolution lines (in 100 Ma steps) of depleted mantle
978 (solid circles) and younger upper continental crust (open circles) are also shown up to 1200 Ma
979 (after Kramers and Tolstikhin, 1997). Note that the isochron projects towards a composition
980 approximately mid-way on the tie-line (stippled) between 1.2 Ga mantle and juvenile crust.

981

982 Fig. 10. Inboard (southeast) – outboard (northwest) schematic view of basin evolution during
983 deposition of the Arctic Bay Fm., based on field lithostratigraphy. Southeastern one-third of the
984 MIG is not shown. (1) During deposition of the lowermost Arctic Bay Fm., the MIG deepened
985 northwestward, the sea-floor was above storm wave-base everywhere, and sediment consisted of
986 quartz sand and silt. At graben margins (not shown), wedges of coarse terrigenous debris began
987 to be deposited (Fabricius Fiord Fm.). (2) The sea-floor deepened and the northwestern MIG was
988 dominated by silt deposited below storm wave-base. Carbonate layers were deposited,
989 diagenetically altered in the subsurface, and locally redeposited as intraclast rudstone. (3) After
990 continued deepening and development of a chemocline, the MIG northwest of Alpha River
991 became a black shale basin. The AR location may also have experienced black shale deposition
992 during intervals recorded by sedimentary rocks now obscured by cover. Geochemical data
993 indicate that the clay was supplied by weathering of exposed basalt (Nauyat Fm.) (4) During
994 deposition of the upper part of the black shale across the central and western MIG, millimetric
995 discoidal calcite concretions grew below the sediment-water interface. Shale Valley was near a
996 locus of deep-water mound development (Ikpiarjuk Fm.), and sooty carbonate, slumps and
997 concretion-clast debrites near the top of the formation are related lithofacies. Syndepositional
998 fault activity is indicated throughout black shale deposition by ITSs at this locality. (5) Abrupt
999 but conformable transition to terrigenous-free dolostone laminite of the Nanisivik Fm. reflects
1000 regional tectonic adjustment of the basin and elimination of clay supply (see Turner, 2004 and
1001 Turner, 2009). Fabricius Fiord Fm. continues to be deposited at graben margins (not shown;
1002 Jackson and Iannelli, 1981).

1003 TABLE CAPTIONS

1004 Table 1. Major element, four-step loss-on-ignition, RockEval, trace element, and Pb-isotope

1005 data.

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Table 1: Major element, total organic carbon and loss on ignition (wt%), trace element (ppb) and

	09-SVC-51	09-SVC-53	09-SVC-55	09-SVC-58	09-SVC-60	09-SVC-62
Height (m)	5.5	12	23	31.5	44	66
Type	FeMn nodule	Rusty sst	FeMn nodule	Shale (calc)	Carb nod	Rusty shale
SiO ₂	31.37	63.96	42.67	59.34		54.56
TiO ₂	0.22	0.3	0.37	0.86		0.73
Al ₂ O ₃	6.12	9.1	10.23	19.78		16.23
Fe ₂ O ₃	33.23	10.66	21.51	6.14		4.03
MnO	2.14	0.55	0.89	0.02		0.01
MgO	2.46	2.58	2.48	2.01		1.28
CaO	2.06	3.74	5.89	0.24		0.31
Na ₂ O	0.3	1.33	0.4	0.47		1.25
K ₂ O	0.8	0.64	1.02	4.7		2.89
P ₂ O ₅	1.05	0.01	4.33	0.06		0.07
LOI	20.17	7.29	9.95	6.56		18.45
SO ₃						
TOC	0.55	0.07	0.22	0.14		9.01
Total	99.93	100.15	99.75	100.19		99.81
N ₂ (105°C)	0.39	0.23	0.51	1.73		2.11
O ₂ (371°C)	0.4	0.22	0.66	0.55		1.5
O ₂ (500°C)	0.52	0.12	0.59	0.26		11.43
O ₂ (1000°C)	16.78	6.11	7.45	3.68		2.12
Total LOI	18.09	6.68	9.21	6.22		17.16
Li	33360	56510	82930	50760	47510	52220
Be	1454	2927	1763	4119	453	2405
Sc	4442	13130	11820	13620	6255	14520
Ti	875400	4944000	3613000	5221000	963700	4258000
V	23140	66300	50400	94440	17730	134000
Cr	14160	62910	42940	76190	16400	71710
Co	9516	27670	18470	11520	26150	2041
Ni	12970	28810	32160	32040	36290	45800
Cu	6445	10790	68340	bdl	1995	25330
Zn	45280	67960	1040000	53140	203300	18840
Ga	6952	23140	17550	28010	6907	23400
As	468	5165	1469	554	214	8102
Rb	29410	140300	82680	224600	12920	132600
Sr	34600	88910	116800	97230	101500	103200
Y	31590	45850	170300	33540	26110	31120
Zr	30450	262300	231500	168300	32820	152000
Nb	2816	17300	11850	17700	2722	11010
Mo	223	484	676	207	50	10440
Ag	19	201	51	27	12	424
Cd	16	94	338	57	69	64
In	23	94	475	71	23	33
Sn	133	3482	1453	5034	521	5001

Figure 1 colour

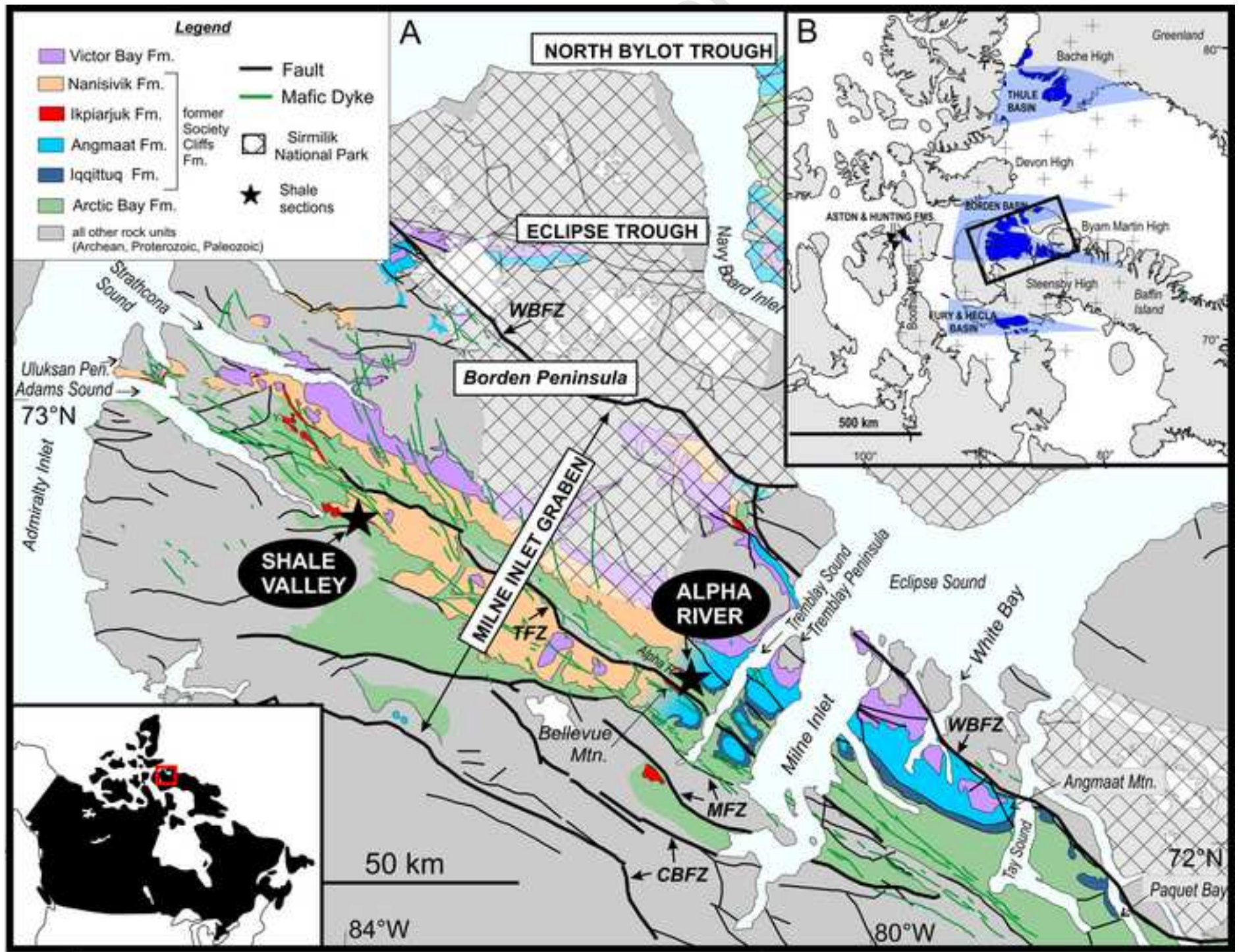


Figure 1 B&W

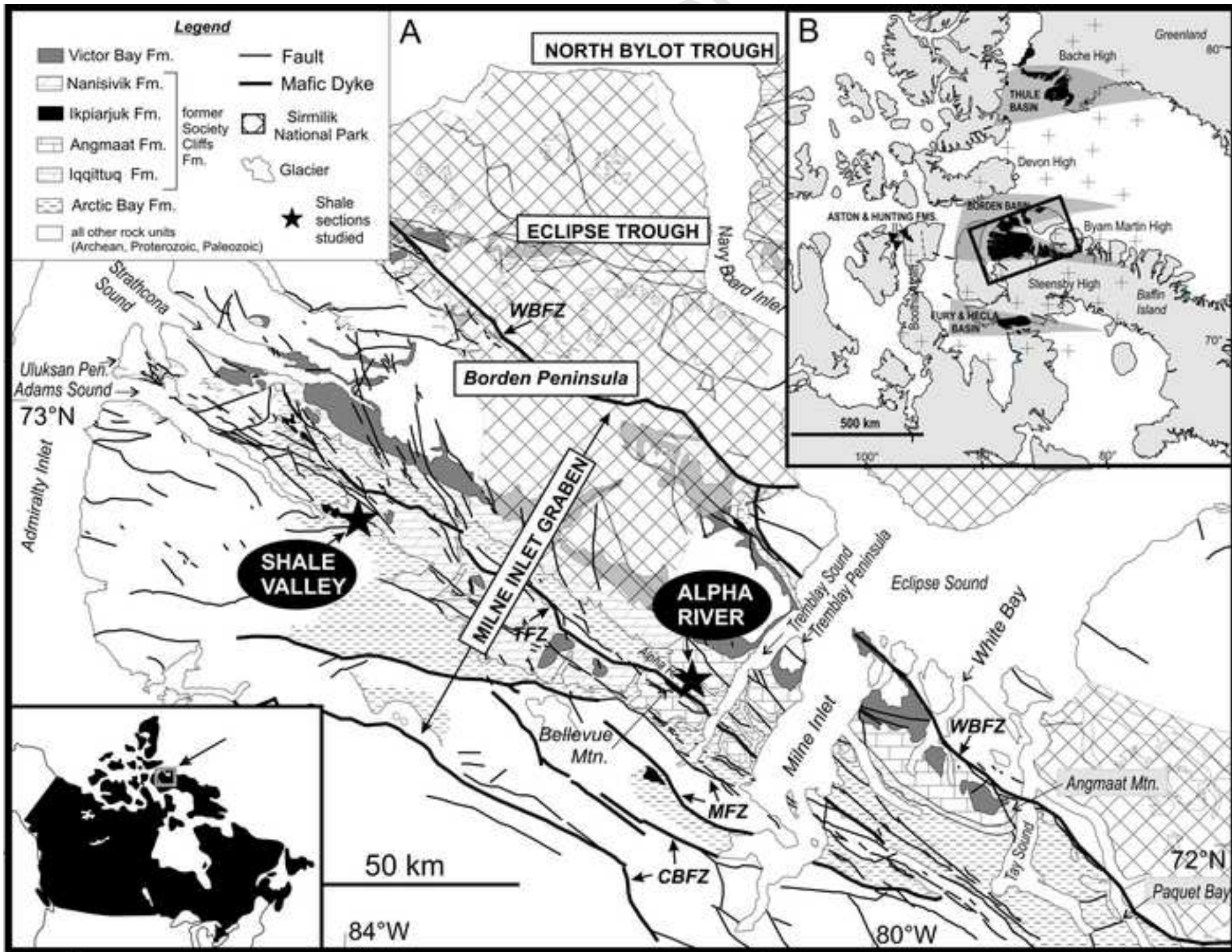


Figure 2

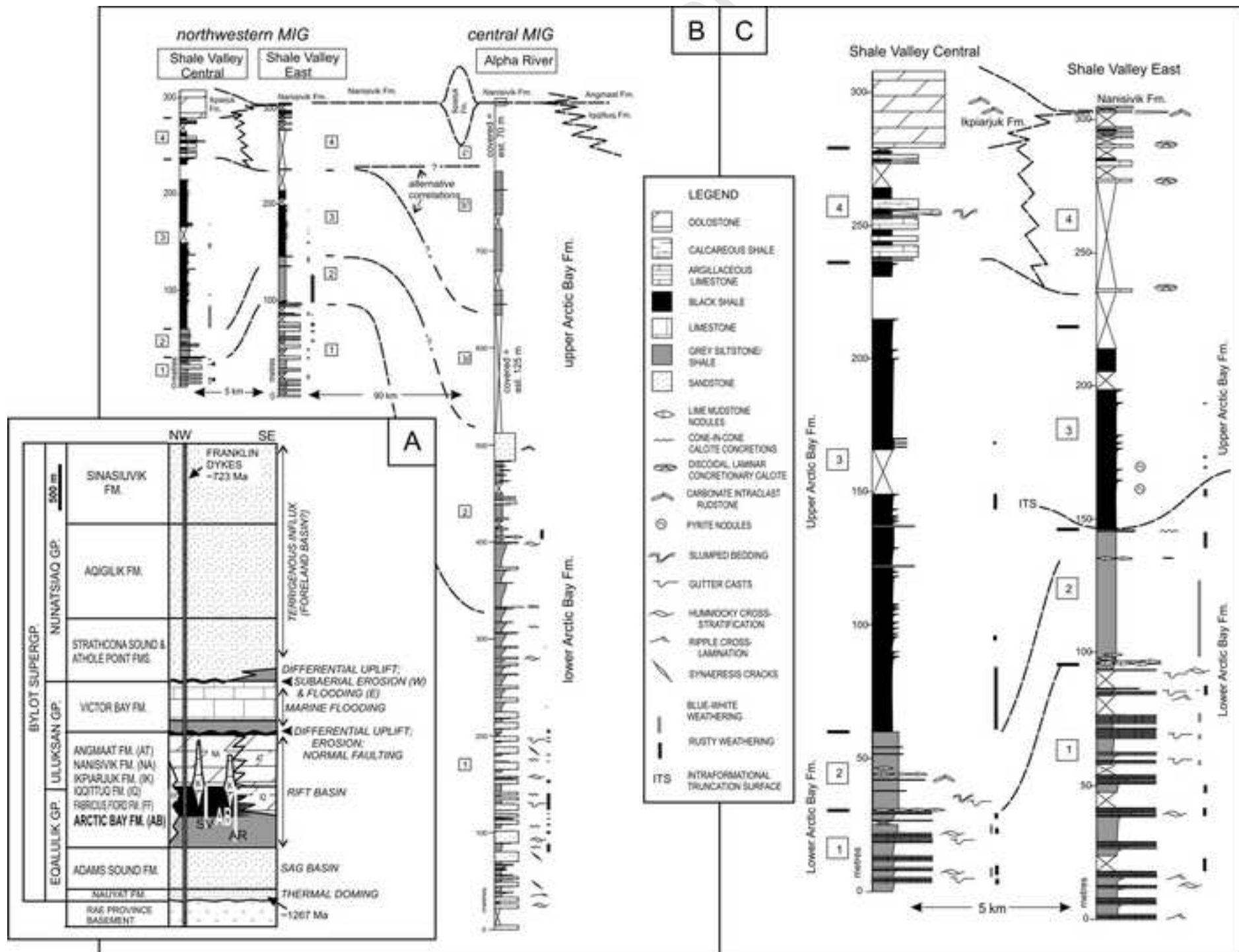


Figure 3 colour

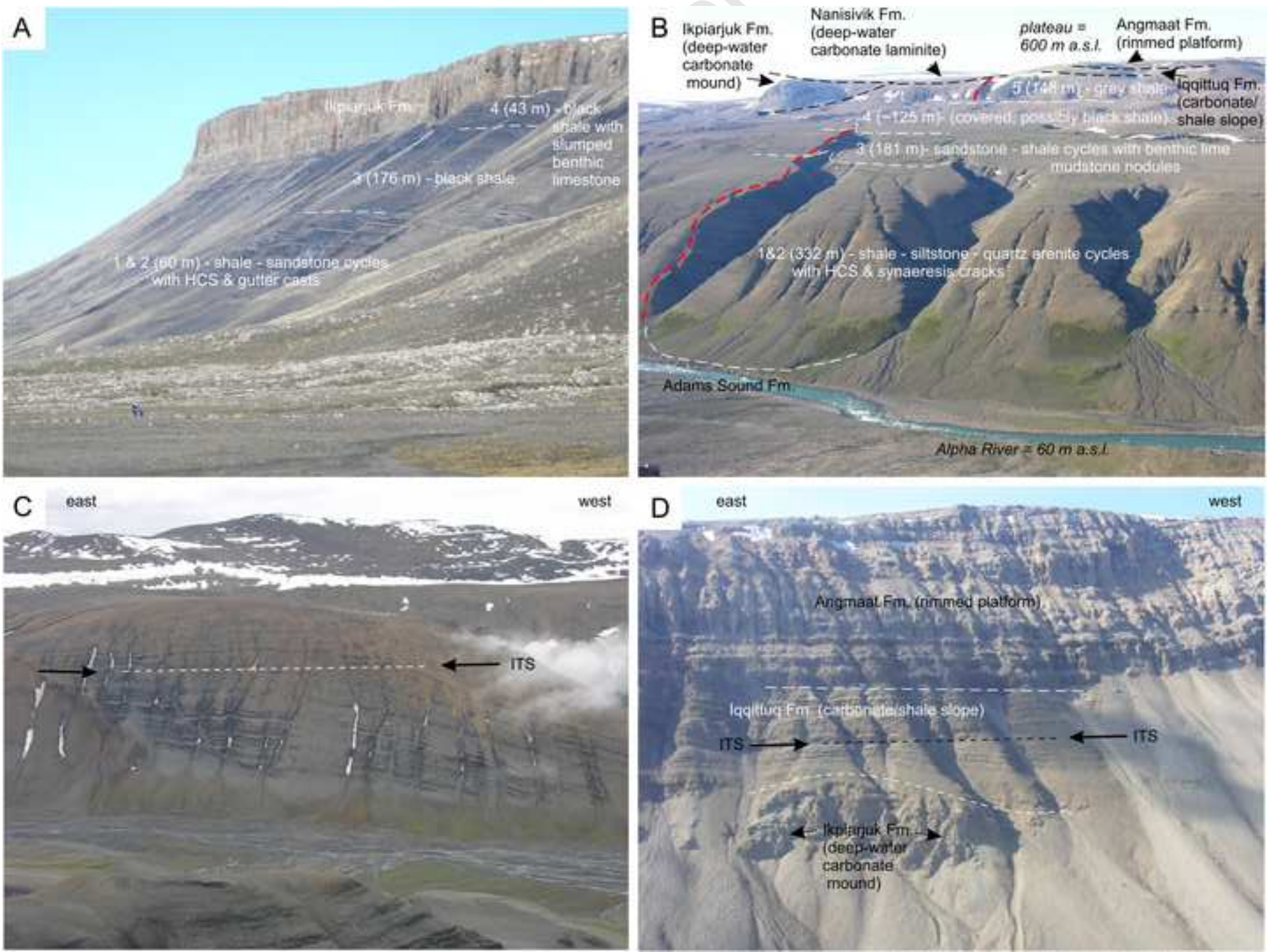
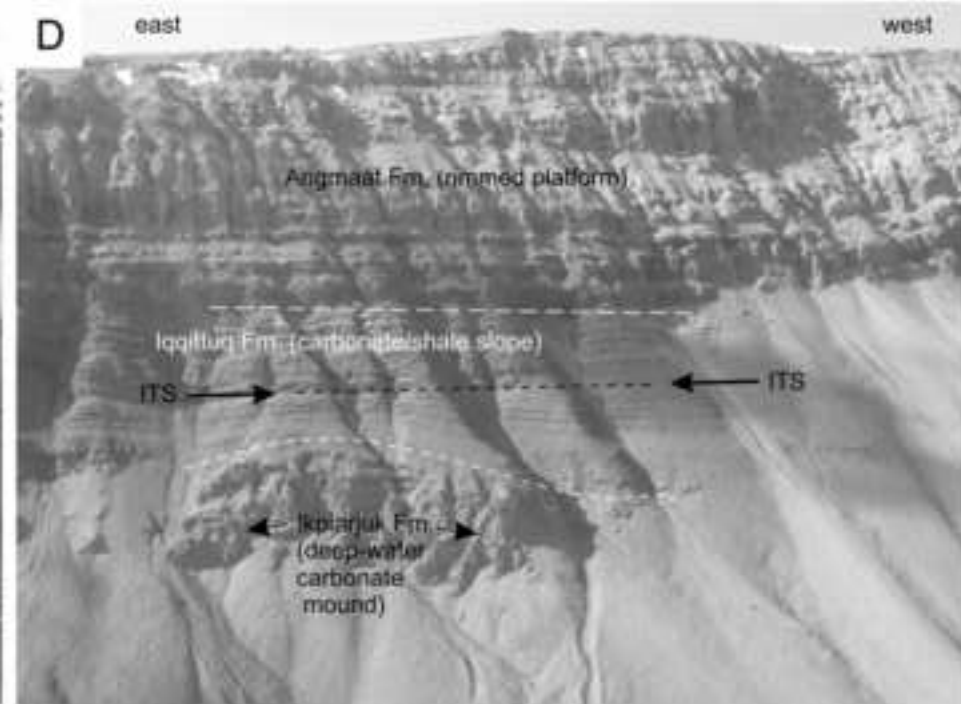
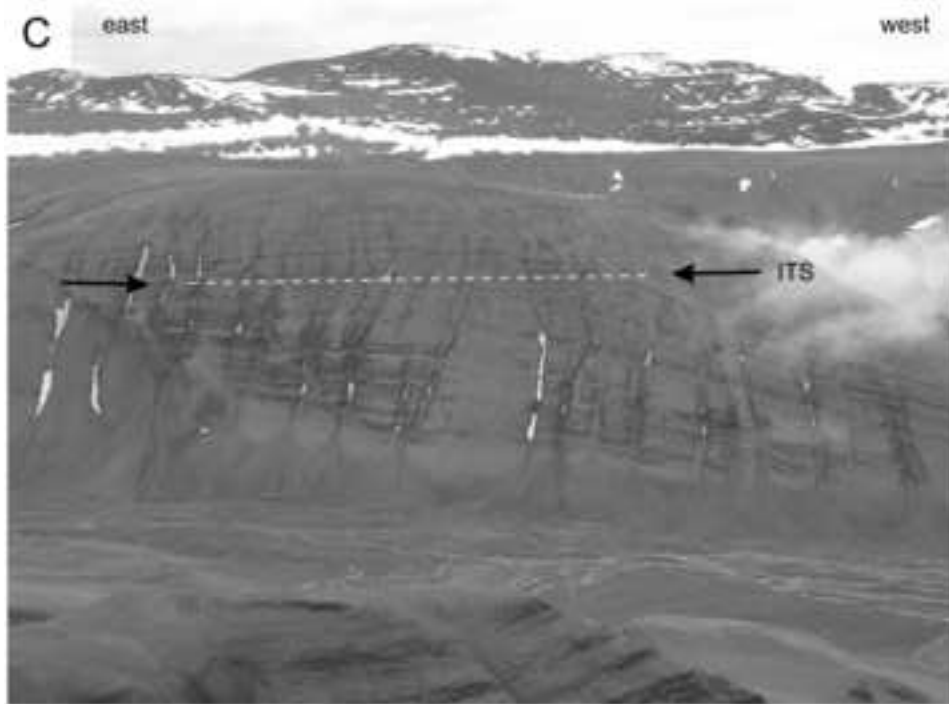
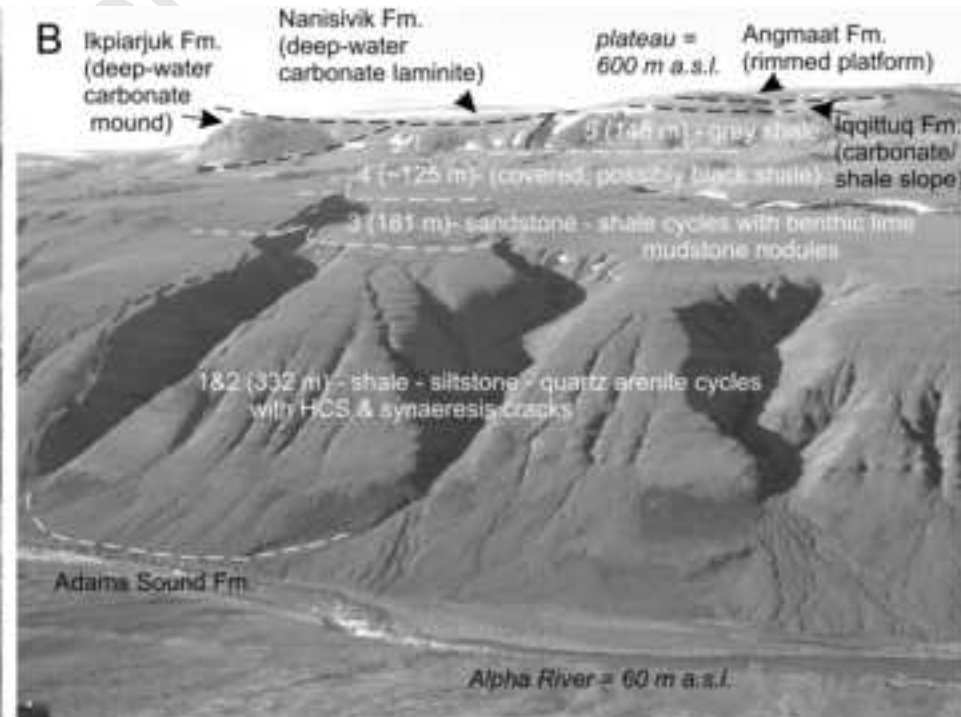
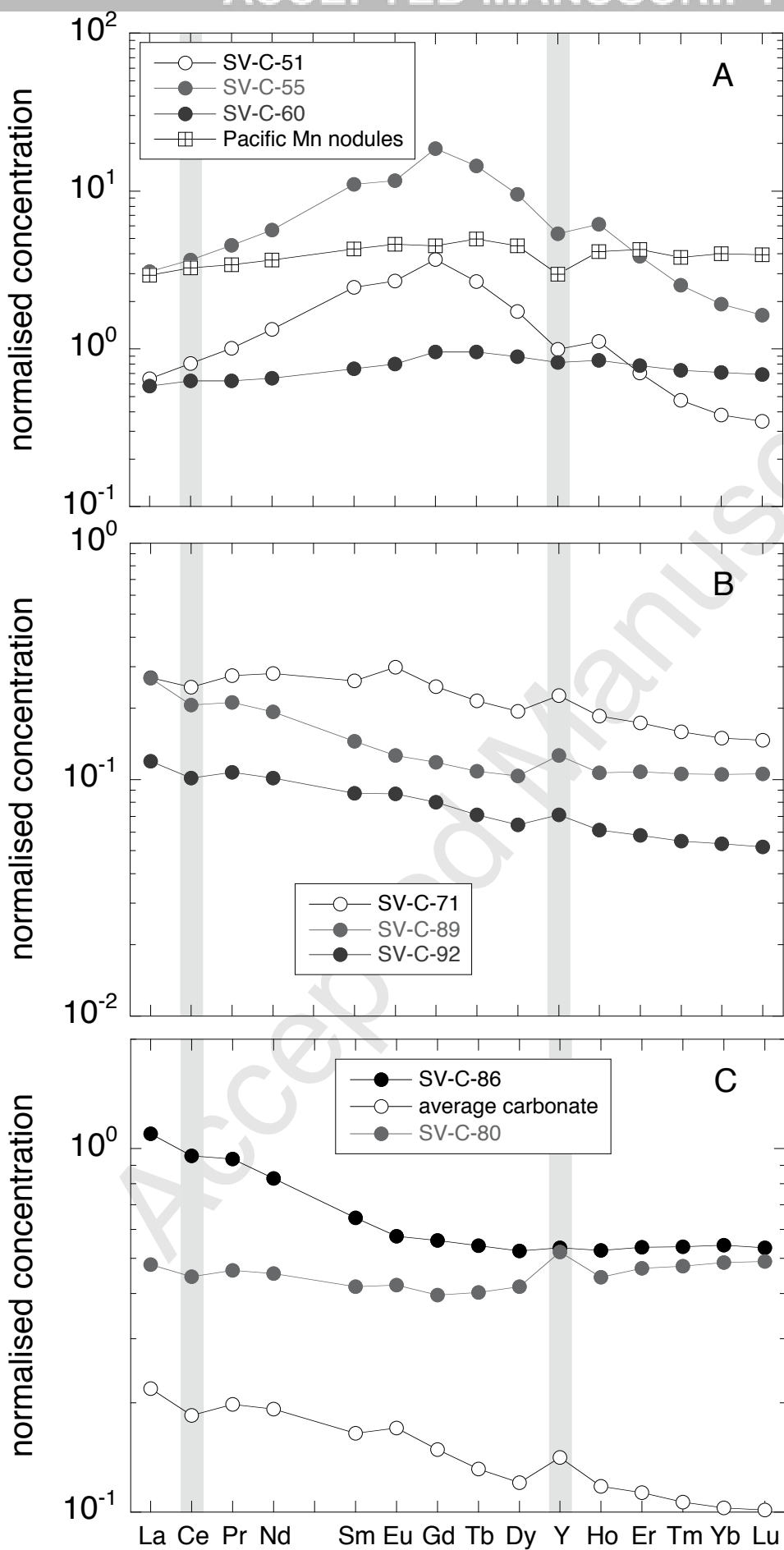
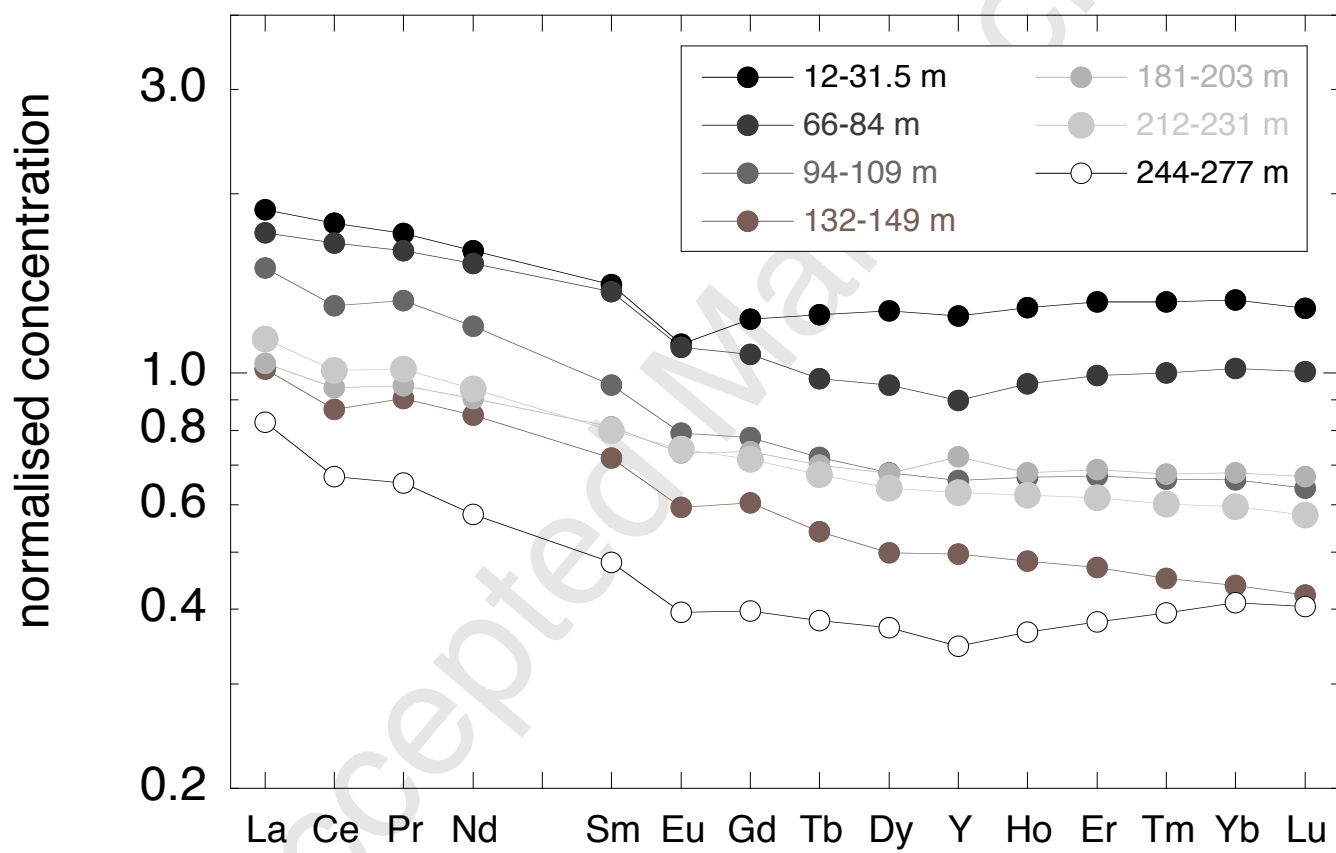


Figure 3 B&W







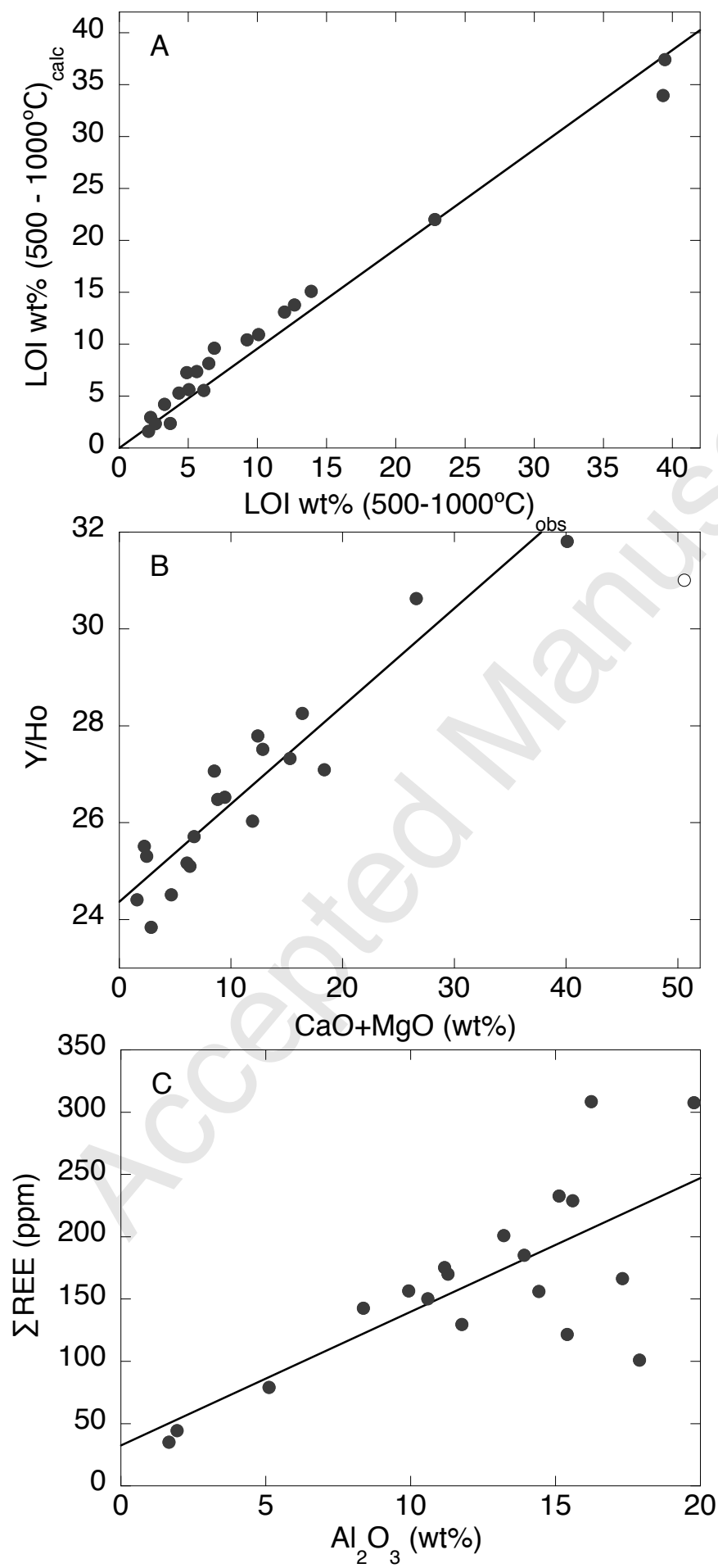


Figure 7

